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**WO 02/08241 A2**

(54) Title: PRODRUGS OF PHOSPHONATE NUCLEOTIDE ANALOGUES AND METHODS FOR SELECTING AND MAKING SAME

(57) Abstract: A novel method is provided for screening prodrugs of methoxyphosphonate nucleotide analogues to identify prodrugs selectively targeting desired tissues with antiviral or antitumor activity. This method has led to the identification of novel mixed ester-amides of PMPA for retroviral or hepadnaviral therapy, including compounds of structure (5a) having substituent groups as defined herein. Compositions of these novel compounds in pharmaceutically acceptable excipients and their use in therapy and prophylaxis are provided. Also provided is an improved method for the use of magnesium alkoxide for the preparation of starting materials and compounds for use herein.

10

## Prodrugs of Phosphonate Nucleotide Analogues and Methods for Selecting and Making Same

This application relates to prodrugs of methoxyphosphonate nucleotide analogues. In particular it relates to improved methods for making and identifying such prodrugs.

Many methoxyphosphonate nucleotide analogues are known. In general, such compounds have the structure A-OCH<sub>2</sub>P(O)(OR)<sub>2</sub> where A is the residue of a nucleoside analogue and R independently is hydrogen or various protecting or prodrug functionalities. See U.S. Patent Nos. 5,663,159, 5,977,061 and 5,798,340, 20 Oliyai et al, "Pharmaceutical Research" 16(11):1687-1693 (1999), Stella et al., "J. Med. Chem." 23(12):1275-1282 (1980), Aarons, L., Boddy, A. and Petrak, K. (1989) *Novel Drug Delivery and Its Therapeutic Application* (Prescott, L. F. and Nimmo, W. S., ed.), pp. 121-126; Bundgaard, H. (1985) *Design of Prodrugs* (Bundgaard, H., ed.) pp. 70-74 and 79-92; Banerjee, P. K. and Amidon, G. L. (1985) *Design of Prodrugs* 25 (Bundgaard, H., ed.) pp. 118-121; Notari, R. E. (1985) *Design of Prodrugs* (Bundgaard, H., ed.) pp. 135-156; Stella, V. J. and Himmelstein, K. J. (1985) *Design of Prodrugs* (Bundgaard, H., ed.) pp. 177-198; Jones, G. (1985) *Design of Prodrugs* (Bundgaard, H., ed.) pp. 199-241; Connors, T. A. (1985) *Design of Prodrugs* (Bundgaard, H., ed.) pp. 291-316. All literature and patent citations herein are 30 expressly incorporated by reference.

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Summary of the Invention

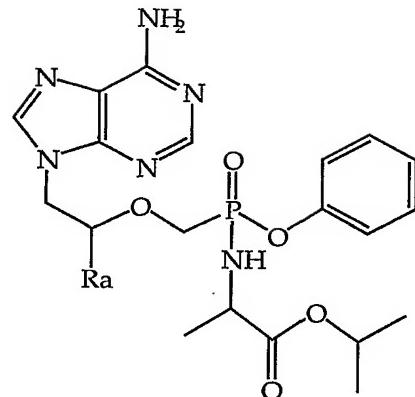
Prodrugs of methoxyphosphonate nucleotide analogues intended for antiviral or antitumor therapy, while known, traditionally have been selected for their systemic effect. For example, such prodrugs have been selected for enhanced bioavailability, i.e., ability to be absorbed from the gastrointestinal tract and converted rapidly to parent drug to ensure that the parent drug is available to all tissues. However, applicants now have found that it is possible to select prodrugs that become enriched at therapeutic sites, as illustrated by the studies described herein where the analogues are enriched at localized focal sites of HIV infection. The objective of this invention is, among other advantages, to produce less toxicity to bystander tissues and greater potency of the parental drug in tissues which are the targets of therapy with the parent methoxyphosphonate nucleotide analogue.

Accordingly, pursuant to these observations, a screening method is provided for identifying a methoxyphosphonate nucleotide analogue prodrug conferring enhanced activity in a target tissue comprising:

- 20 (a) providing at least one of said prodrugs;
- (b) selecting at least one therapeutic target tissue and at least one non-target tissue;
- (c) administering the prodrug to the target tissue and to said at least one non-target tissue; and
- 25 (d) determining the relative antiviral activity conferred by the prodrug in the tissues in step (c).

In preferred embodiments, the target tissue are sites where HIV is actively replicated and/or which serve as an HIV reservoir, and the non-target tissue is an intact animal. Unexpectedly, we found that selecting lymphoid tissue as the target 30 tissue for the practice of this method for HIV led to identification of prodrugs that enhance the delivery of active drug to such tissues.

A preferred compound of this invention, which has been identified by this method has the structure (1),



5

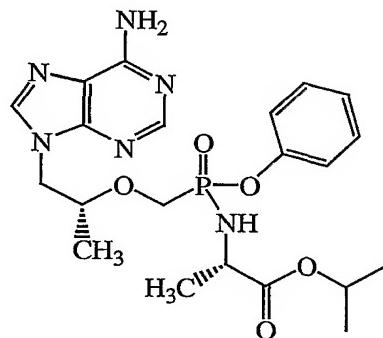
(1)

10 where Ra is H or methyl,

and chirally enriched compositions thereof, salts, their free base and solvates thereof.

A preferred compound of this invention has the structure (2)

15



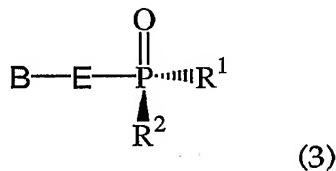
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(2)

and its enriched diasteromers, salts, free base and solvates.

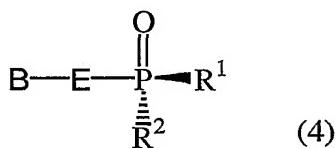
25 In addition, we unexpectedly found that the chirality of substituents on the phosphorous atom and/or the amide substituent are influential in the enrichment observed in the practice of this invention. Thus, in another embodiment of this

5 invention, we provide diastereomerically enriched compounds of this invention  
having the structure (3)



which are substantially free of the diastereomer (4)

15



20 wherein

$R^1$  is an oxyester which is hydrolyzable *in vivo*, or hydroxyl;

B is a heterocyclic base;

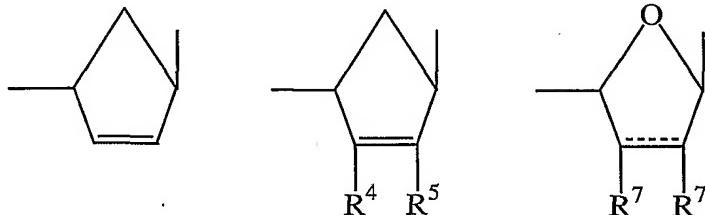
$R^2$  is hydroxyl, or the residue of an amino acid bonded to the P atom

through an amino group of the amino acid and having each carboxy substituent of

25 the amino acid optionally esterified, but not both of  $R^1$  and  $R^2$  are hydroxyl;

E is  $-(CH_2)_2-$ ,  $-CH(CH_3)CH_2-$ ,  $-CH(CH_2F)CH_2-$ ,  $-CH(CH_2OH)CH_2-$ ,  
 $-CH(CH=CH_2)CH_2-$ ,  $-CH(C\equiv CH)CH_2-$ ,  $-CH(CH_2N_3)CH_2-$ ,

30



$-CH(R^6)OCH(R^{6'})-$ ,  $-CH(R^9)CH_2O-$  or  $-CH(R^8)O-$ , wherein the right hand bond is

35 linked to the heterocyclic base;

the broken line represents an optional double bond;

$R^4$  and  $R^5$  are independently hydrogen, hydroxy, halo, amino or a  
substituent having 1-5 carbon atoms selected from acyloxy, alkyoxy, alkylthio,  
alkylamino and dialkylamino;

5        R<sup>6</sup> and R<sup>6'</sup> are independently H, C<sub>1</sub>-C<sub>6</sub> alkyl, C<sub>1</sub>-C<sub>6</sub> hydroxyalkyl, or C<sub>2</sub>-C<sub>7</sub> alkanoyl;

R<sup>7</sup> is independently H, C<sub>1</sub>-C<sub>6</sub> alkyl, or are taken together to form -O- or -CH<sub>2</sub>-;

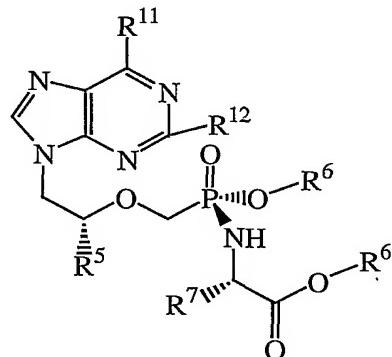
R<sup>8</sup> is H, C<sub>1</sub>-C<sub>6</sub> alkyl, C<sub>1</sub>-C<sub>6</sub> hydroxyalkyl or C<sub>1</sub>-C<sub>6</sub> haloalkyl; and

10      R<sup>9</sup> is H, hydroxymethyl or acyloxymethyl;

and their salts, free base, and solvates.

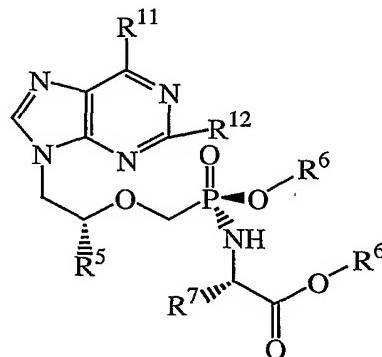
The diastereomers of structure (3) are designated the (S) isomers at the phosphorus chiral center.

Preferred embodiments of this invention are the diastereomerically enriched  
15      compounds having the structure (5a)



(5a)

25      which is substantially free of diastereomer (5b)



(5b)

5 wherein

$R^5$  is methyl or hydrogen;

10  $R^6$  independently is H, alkyl, alkenyl, alkynyl, aryl or arylalkyl, or  $R^6$  independently is alkyl, alkenyl, alkynyl, aryl or arylalkyl which is substituted with from 1 to 3 substituents selected from alkylamino, alkylaminoalkyl, dialkylaminoalkyl, dialkylamino, hydroxyl, oxo, halo, amino, alkylthio, alkoxy, alkoxyalkyl, aryloxy, aryloxyalkyl, arylalkoxy, arylalkoxyalkyl, haloalkyl, nitro, nitroalkyl, azido, azidoalkyl, alkylacyl, alkylacylalkyl, carboxyl, or alkylacylamino;

15  $R^7$  is the side chain of any naturally-occurring or pharmaceutically acceptable amino acid and which, if the side chain comprises carboxyl, the carboxyl group is optionally esterified with an alkyl or aryl group;

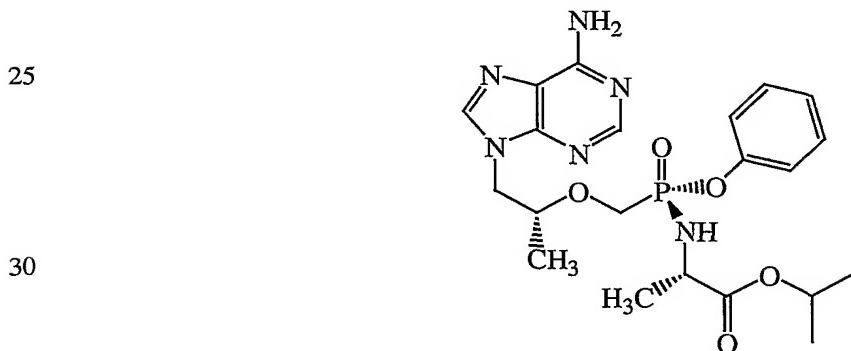
$R^{11}$  is amino, alkylamino, oxo, or dialkylamino; and

$R^{12}$  is amino or H;

and its salts, tautomers, free base and solvates.

20 A preferred embodiment of this invention is the compound of structure (6),

9-[(*R*)-2-[[*(S*)-[[*(S*)-1-(isopropoxycarbonyl)ethyl]amino]phenoxyphosphinyl]methoxylpropyl]adenine, also designated herein GS-7340

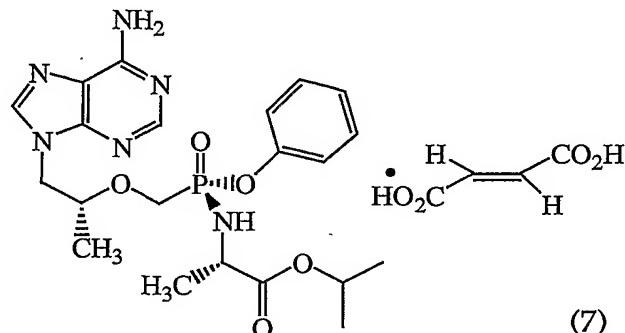


(6)

35 Another preferred embodiment of this invention is the fumarate salt of structure (5) (structure (7)), 9-[(*R*)-2-[[*(S*)-[[*(S*)-1-

- 5 (isopropoxycarbonyl)ethyl]aminolphenoxyphosphinyl]methoxylpropyl]adenine  
fumarate (1:1), also designated herein GS-7340-2

10



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The compounds of structures (1)-(7) optionally are formulated into compositions containing pharmaceutically acceptable excipients. Such compositions are used in effective doses in the therapy or prophylaxis of viral (particularly HIV or hepadnaviral) infections.

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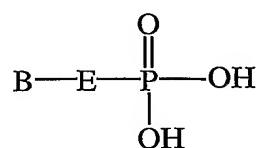
In a further embodiment, a method is provided for the facile manufacture of 9-[2-(phosphonomethoxy)propyl]adenine (hereinafter "PMPA" or 9-[2-(phosphonomethoxy)ethyl] adenine (hereinafter "PMEA") using magnesium alkoxide, which comprises combining 9-(2-hydroxypropyl)adenine or 9-(2-hydroxyethyl)adenine, protected *p*-toluenesulfonyloxymethylphosphonate and magnesium alkoxide, and recovering PMPA or PMEA, respectively.

25

#### Detailed Description of the Invention

The methoxyphosphonate nucleotide analogue parent drugs for use in this screening method are compounds having the structure A-OH<sub>2</sub>P(O)(OH)<sub>2</sub> wherein 30 A is the residue of a nucleoside analogue. These compounds are known per se and are not part of this invention. More particularly, the parent compounds comprise a heterocyclic base B and an aglycon E, in general having the structure

35



5 wherein the group B is defined below and group E is defined above. Examples are described in U.S. Patent Nos. 4,659,825, 4,808,716, 4,724,233, 5,142,051, 5,130,427, 5,650,510, 5,663,159, 5,302,585, 5,476,938, 5,696,263, 5,744,600, 5,688,778, 5,386,030, 5,733,896, 5,352,786, and 5,798,340, and EP 821,690 and 654,037.

The prodrugs for use in the screening method of this invention are  
10 covalently modified analogues of the parent methoxyphosphonate nucleotide analogues described in the preceding paragraph. In general, the phosphorus atom of the parent drug is the preferred site for prodrug modification, but other sites are found on the heterocyclic base B or the aglycon E. Many such prodrugs are already known. Primarily, they are esters or amides of the phosphorus atom, but also  
15 include substitutions on the base and aglycon. None of these modifications per se is part of this invention and none are to be considered limiting on the scope of the invention herein.

The phosphorus atom of the methoxyphosphonate nucleotide analogues contains two valences for covalent modification such as amidation or esterification  
20 (unless one phosphoryl hydroxyl is esterified to an aglycon E hydroxyl substituent, whereupon only one phosphorus valence is free for substitution). The esters typically are aryloxy. The amides ordinarily are naturally occurring monoamino acids having free carboxyl group(s) esterified with an alkyl or aryl group, usually phenyl, cycloalkyl, or *t*-, *n*- or *s*- alkyl groups. Suitable prodrugs for use in the  
25 screening method of this invention are disclosed for example in U.S. Patent No. 5,798,340. However, any prodrug which is potentially believed to be capable of being converted *in vivo* within target tissue cells to the free methoxyphosphonate nucleotide analogue parent drug, e.g., whether by hydrolysis, oxidation, or other covalent transformation resulting from exposure to biological tissues, is suitable for  
30 use in the method of this invention. Such prodrugs may not be known at this time but are identified in the future and thus become suitable candidates available for testing in the method of this invention. Since the prodrugs are simply candidates for screening in the methods their structures are not relevant to practicing or enabling the screening method, although of course their structures ultimately are  
35 dispositive of whether or not a prodrug will be shown to be selective in the assay.

5        The pro-moieties bound to the parent drug may be the same or different. However, each prodrug to be used in the screening assay will differ structurally from the other prodrugs to be tested. Distinct, i.e. structurally different, prodrugs generally are selected on the basis of either their stereochemistry or their covalent structure, or these features are varied in combination. Each prodrug tested,  
10 however, desirably is structurally and stereochemically substantially pure, else the output of the screening assay will be less useful. It is of course within the scope of this invention to test only a single prodrug in an individual embodiment of the method of this invention, although typically then one would compare the results with prior studies with other prodrugs.

15      We have found that the stereochemistry of the prodrugs is capable of influencing the enrichment in target tissues. Chiral sites are at the phosphorus atom and are also found in its substituents. For example, amino acid used in preparing amidates may be D or L forms, and the phosphonate esters or the amino acid esters can contain chiral centers as well. Chiral sites also are found on the  
20 nucleoside analogue portion of the molecules, but these typically are already dictated by the stereochemistry of the parent drug and will not be varied as part of the screen. For example the R isomer of PMPA is preferred as it is more active than the corresponding S isomer. Typically these diasteromers or enantiomers will be chirally enriched if not pure at each site so that the results of the screen will be  
25 more meaningful. As noted, distinctiveness of stereoisomers is conferred by enriching or purifying the stereoisomer (typically this will be a diastereomer rather than an enantiomer in the case of most methoxyphosphonate nucleotide analogues) free of other stereoisomers at the chiral center in question, so that each test compound is substantially homogeneous. By substantially homogeneous or  
30 chirally enriched, we mean that the desired stereoisomer constitutes greater than about 60% by weight of the compound, ordinarily greater than about 80% and preferably greater than about 95%.

Novel Screening Method

Once at least one candidate prodrug has been selected, the remaining steps of the screening method of this invention are used to identify a prodrug possessing the required selectivity for the target tissue. Most conveniently the prodrugs are labeled with a detectable group, e.g. radiolabeled, in order to facilitate detection later in tissues or cells. However, a label is not required since other suitable assays for the prodrug or its metabolites (including the parent drug) can also be employed. These assays could include mass spectrometry, HPLC, bioassays or immunoassays for instance. The assay may detect the prodrug and any one or more of its metabolites, but preferably the assay is conducted to detect only the generation of the parent drug. This is based on the assumption (which may not be warranted in all cases) that the degree and rate of conversion of prodrug to antivirally active parent diphosphate is the same across all tissues tested.

Otherwise, one can test for the diphosphate.

The target tissue preferably will be lymphoid tissue when screening for prodrugs useful in the treatment of HIV infection. Lymphoid tissue will be known to the artisan and includes CD4 cells, lymphocytes, lymph nodes, macrophages and macrophage-like cells including monocytes such as peripheral blood monocytic cells (PBMCs) and glial cells. Lymphoid tissue also includes non-lymphoid tissues that are enriched in lymphoid tissues or cells, e.g. lung, skin and spleen. Other targets for other antiviral drugs of course will be the primary sites of replication or latency for the particular virus concerned, e.g., liver for hepatitis and peripheral nerves for HSV. Similarly, target tissues for tumors will in fact be the tumors themselves. These tissues are all well-known to the artisan and would not require undue experimentation to select. When screening for antiviral compounds, target tissue can be infected by the virus.

Non-target tissues or cells also are screened as part of the method herein. Any number or identity of such tissues or cells can be employed in this regard. In general, tissues for which the parent drug is expected to be toxic will be used as non-target tissues. The selection of a non-target tissue is entirely dependent upon

5 the nature of the prodrug and the activity of the parent. For example, non-hepatic tissues would be selected for prodrugs against hepatitis, and untransformed cells of the same tissue as the tumor will suffice for the antitumor-selective prodrug screen.

It should be noted that the method of this invention is distinct from studies typically undertaken to determine oral bioavailability of prodrugs. In oral  
10 bioavailability studies, the objective is to identify a prodrug which passes into the systemic circulation substantially converted to parent drug. In the present invention, the objective is to find prodrugs that are not metabolized in the gastrointestinal tract or circulation. Thus, target tissues to be evaluated in the method of this invention generally do not include the small intestines or, if the  
15 intestines are included, then the tissues also include additional tissues other than the small intestines.

The target and non-target tissues used in the screening method of this invention typically will be in an intact living animal. Prodrugs containing esters are more desirably tested in dogs, monkeys or other animals than rodents; mice  
20 and rat plasma contains high circulating levels of esterases that may produce a misleading result if the desired therapeutic subject is a human or higher mammal.

It is not necessary to practice this method with intact animals. It also is within the scope of this invention to employ perfused organs, *in vitro* culture of organs (e.g. skin grafts) or cell lines maintained in various forms of cell culture, e.g.  
25 roller bottles or zero gravity suspension systems. For example, MT-2 cells can be used as a target tissue for selecting HIV prodrugs. Thus, the term "tissue" shall not be construed to require organized cellular structures, or the structures of tissues as they may be found in nature, although such would be preferred. Rather, the term "tissue" shall be construed to be synonymous with cells of a particular source,  
30 origin or differentiation stage.

The target and non-target tissue may in fact be the same tissue, but the tissues will be in different biological status. For example, the method herein could be used to select for prodrugs that confer activity in virally-infected tissue (target tissue) but which remain substantially inactive in virally-uninfected cells  
35 (corresponding non-target tissue). The same strategy would be employed to select

5 prophylactic prodrugs, i.e., prodrugs metabolized to antivirally active forms  
incidental to viral infection but which remain substantially unmetabolized in  
uninfected cells. Similarly, prodrugs could be screened in transformed cells and  
the untransformed counterpart tissue. This would be particularly useful in  
comparative testing to select prodrugs for the treatment of hematological  
10 malignancies, e.g. leukemias.

Without being limited by any particular theory of operation, tissue selective  
prodrugs are thought to be selectively taken up by target cells and/or selectively  
metabolized within the cell, as compared to other tissues or cells. The unique  
15 advantage of the methoxyphosphonate prodrugs herein is that their metabolism to  
the dianion at physiological pH ensures that they will be unable to diffuse back out  
of the cell. They therefore remain effective for lengthy periods of time and are  
maintained at elevated intracellular concentrations, thereby exhibiting increased  
potency. The mechanisms for enhanced activity in the target tissue are believed to  
20 include enhanced uptake by the target cells, enhanced intracellular retention, or  
both mechanisms working together. However, the manner in which selectivity or  
enhanced delivery occurs in the target tissue is not important. It also is not  
important that all of the metabolic conversion of the prodrug to the parent  
compound occurs within the target tissue. Only the final drug activity-conferring  
conversion need occur in the target tissue; metabolism in other tissues may provide  
25 intermediates finally converted to antiviral forms in the target tissue.

The degree of selectivity or enhanced delivery that is desired will vary with  
the parent compound and the manner in which it is measured (% dose distribution  
or parent drug concentration). In general, if the parent drug already possess a  
generous therapeutic window, a low degree of selectivity may be sufficient for the  
30 desired prodrug. On the other hand, toxic compounds may require more extensive  
screening to identify selective prodrugs. The relative expense of the method of this  
invention can be reduced by screening only in the target tissue and tissues against  
which the parent compound is known to be relatively toxic, e.g. for PMEA, which  
is nephrotoxic at higher doses, the primary focus will be on kidney and lymphoid  
35 tissues.

5        The step of determining the relative antiviral activity of a prodrug in the selected tissues ordinarily is accomplished by assaying target and non-target tissues for the relative presence or activity of a metabolite of the prodrug, which metabolite is known to have, or is converted to, a metabolite having antiviral or antitumor activity. Thus, typically one would determine the relative amount of the  
10      parent drug in the tissues over substantially the same time course in order to identify prodrugs that are preferentially metabolized in the target tissue to an antivirally or antitumor active metabolite or precursor thereof which in the target tissue ultimately produces the active metabolite. In the case of antiviral compounds, the active metabolite is the diphosphate of the phosphonate parent  
15      compounds. It is this metabolite that is incorporated into the viral nucleic acid, thereby truncating the elongating nucleic acid strand and halting viral replication. Metabolites of the prodrug can be anabolic metabolites, catabolic metabolites, or the product of anabolism and cabolism together. The manner in which the metabolite is produced is not important in the practice of the method of this  
20      invention.

      The method of this invention is not limited to assaying a metabolite which per se possesses antiviral or antitumor activity. Instead, one can assay inactive precursors of the active metabolites. Precursors of the antivirally active diphosphate metabolite include the monophosphate of the parent drug, monophosphates of other metabolites of the parent drug (e.g., an intermediate modification of a substituent on the heterocyclic base), the parent itself and metabolites generated by the cell in converting the prodrug to the parent prior to phosphorylation. The precursor structures may vary considerably as they are the result of cellular metabolism. However, this information is already known or could  
25      be readily determined by one skilled in the art.  
30     

      If the prodrug being assayed does not exhibit antitumor or antiviral activity per se then adjustments to the raw assay results may be required. For example, if the intracellular processing of the inactive metabolite to an active metabolite occurs at different rates among the tissues being tested, the raw assay results with the  
35      inactive metabolite would need to be adjusted to take account of the differences

5 among the cell types because the relevant parameter is the generation of activity in the target tissue, not accumulation of inactive metabolites. However, determining the proper adjustments would be within the ordinary skill. Thus, when step (d) of the method herein calls for determining the activity, activity can be either measured directly or extrapolated. It does not mean that the method herein is  
10 limited to only assaying intermediates that are active per se. For instance, the absence or decline of the prodrug in the test tissues also could be assayed. Step (d) only requires assessment of the activity conferred by the prodrug as it interacts with the tissue concerned, and this may be based on extrapolation or other indirect measurement.

15 Step (d) of the method of this invention calls for determining the "relative" activity of the prodrug. It will be understood that this does not require that each and every assay or series of assays necessarily must also contain runs with the selected non-target tissue. On the contrary, it is within the scope of this invention to employ historical controls of the non-target tissue or tissues, or algorithms  
20 representing results to be expected from such non-target tissues, in order to provide the benchmark non-target activity.

The results obtained in step (d) are then used optimally to select or identify a prodrug which produces greater antiviral activity in the target tissue than in the non-target tissue. It is this prodrug that is selected for further development.

25 It will be appreciated that some preassessment of prodrug candidates can be undertaken before the practice of the method of this invention. For example, the prodrug will need to be capable of passing largely unmetabolized through the gastrointestinal tract, it will need to be substantially stable in blood, and it should be able to permeate cells at least to some degree. In most cases it also will need to  
30 complete a first pass of the hepatic circulation without substantial metabolism. Such prestudies are optional, and are well-known to those skilled in the art.

The same reasoning as is described above for antiviral activity is applicable to antitumor prodrugs of methoxyphosphonate nucleotide analogues as well. These include, for example, prodrugs of PMEG, the guanyl analogue of PMEA. In

5 this case, cytotoxic phosphonates such as PMEG are worthwhile candidates to pursue as their cytotoxicity in fact confers their antitumor activity.

A compound identified by this novel screening method then can be entered into a traditional preclinical or clinical program to confirm that the desired objectives have been met. Typically, a prodrug is considered to be selective if the 10 activity or concentration of parent drug in the target tissue (% dose distribution) is greater than 2x, and preferably 5x, that of the parent compound in non-target tissue. Alternatively, a prodrug candidate can be compared against a benchmark prodrug. In this case, selectivity is relative rather than absolute. Selective prodrugs will be those resulting in greater than about 10x concentration or activity in the 15 target tissue as compared with the prototype, although the degree of selectivity is a matter of discretion.

#### Novel Method for Preparation of Starting Materials or Intermediates

Also included herein is an improved method for manufacture of preferred 20 starting materials (parent drugs) of this invention, PMEA and (R)-PMPA.

Typically, this method comprises reacting 9-(2-hydroxypropyl)adenine (HPA) or 9-(2-hydroxyethyl)adenine (HEA) with a magnesium alkoxide, thereafter adding the protected aglycon synthon *p*-toluene-sulfonyloxymethylphosphonate (tosylate) to the reaction mixture, and recovering PMPA or PMEA, respectively.

25 Preferably, HPA is the enriched or isolated R enantiomer. If a chiral HPA mixture is used, R-PMPA can be isolated from the chiral PMPA mixture after the synthesis is completed.

Typically the tosylate is protected by lower alkyl groups, but other suitable 30 groups will be apparent to the artisan. It may be convenient to employ the tosylate presubstituted with the prodrug phosphonate substituents which are capable of acting as protecting groups in the tosylation reaction, thereby allowing one to bypass the deprotection step and directly recover prodrug or an intermediate therefore.

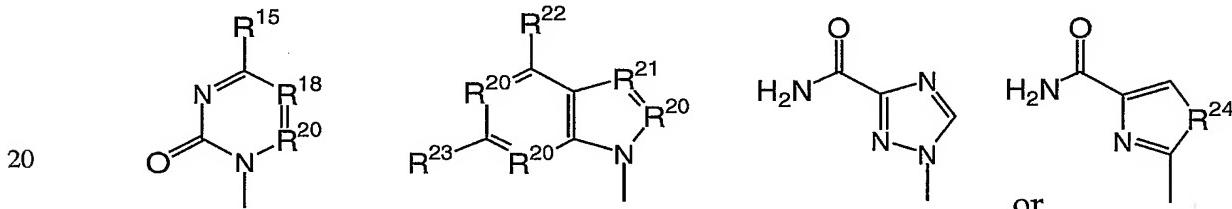
The alkyl group of the magnesium alkoxide is not critical and can be any C<sub>1</sub>- 35 C<sub>6</sub> branched or normal alkyl, but is preferably t-butyl (for PMPA) or isopropyl (for

5 PMEA). The reaction conditions also are not critical, but preferably comprise heating the reaction mixture at about 70-75°C with stirring or other moderate agitation.

If there is no interest in retaining the phosphonate substituents, the product is deprotected (usually with bromotrimethylsilane where the tosylate protecting group is alkyl), and the product then recovered by crystallization or other conventional method as will be apparent to the artisan.

### Heterocyclic Base

In the compounds of this invention depicted in structures (3) and (4), the heterocyclic base B is selected from the structures



wherein

R<sup>15</sup> is H, OH, F, Cl, Br, I, OR<sup>16</sup>, SH, SR<sup>16</sup>, NH<sub>2</sub>, or NHR<sup>17</sup>;

25 R<sup>16</sup> is C<sub>1</sub>-C<sub>6</sub> alkyl or C<sub>2</sub>-C<sub>6</sub> alkenyl including CH<sub>3</sub>, CH<sub>2</sub>CH<sub>3</sub>, CH<sub>2</sub>CCH, CH<sub>2</sub>CHCH<sub>2</sub> and C<sub>3</sub>H<sub>7</sub>;

R<sup>17</sup> is C<sub>1</sub>-C<sub>6</sub> alkyl or C<sub>2</sub>-C<sub>6</sub> alkenyl including CH<sub>3</sub>, CH<sub>2</sub>CH<sub>3</sub>, CH<sub>2</sub>CCH, CH<sub>2</sub>CHCH<sub>2</sub>, and C<sub>3</sub>H<sub>7</sub>;

R<sup>18</sup> is N, CF, CCl, CBr, Cl, CR<sup>19</sup>, CSR<sup>19</sup>, or COR<sup>19</sup>;

30 R<sup>19</sup> is H, C<sub>1</sub>-C<sub>9</sub> alkyl, C<sub>2</sub>-C<sub>9</sub> alkenyl, C<sub>2</sub> - C<sub>9</sub> alkynyl, C<sub>1</sub>-C<sub>9</sub> alkyl-C<sub>1</sub>-C<sub>9</sub> alkoxy, or C<sub>7</sub>-C<sub>9</sub> aryl-alkyl unsubstituted or substituted by OH, F, Cl, Br or I, R<sup>19</sup> therefore including -CH<sub>3</sub>, -CH<sub>2</sub>CH<sub>3</sub>, -CHCH<sub>2</sub>, -CHCHBr, -CH<sub>2</sub>CH<sub>2</sub>Cl, -CH<sub>2</sub>CH<sub>2</sub>F, -CH<sub>2</sub>CCH, -CH<sub>2</sub>CHCH<sub>2</sub>, -C<sub>3</sub>H<sub>7</sub>, -CH<sub>2</sub>OH, -CH<sub>2</sub>OCH<sub>3</sub>, -CH<sub>2</sub>OC<sub>2</sub>H<sub>5</sub>, -CH<sub>2</sub>OCC<sub>3</sub>H<sub>7</sub>, -CH<sub>2</sub>OCH<sub>2</sub>CHCH<sub>2</sub>, -CH<sub>2</sub>C<sub>3</sub>H<sub>7</sub>, -CH<sub>2</sub>CH<sub>2</sub>OH, -CH<sub>2</sub>CH<sub>2</sub>OCH<sub>3</sub>,

- 5    -CH<sub>2</sub>CH<sub>2</sub>OC<sub>2</sub>H<sub>5</sub>, -CH<sub>2</sub>CH<sub>2</sub>OCCH, -CH<sub>2</sub>CH<sub>2</sub>OCH<sub>2</sub>CHCH<sub>2</sub>, and  
   -CH<sub>2</sub>CH<sub>2</sub>OC<sub>3</sub>H<sub>7</sub>;
- R<sup>20</sup> is N or CH;
- R<sup>21</sup> is N, CH, CCN, CCF<sub>3</sub>, CC≡CH or CC(O)NH<sub>2</sub>;
- R<sup>22</sup> is H, OH, NH<sub>2</sub>, SH, SCH<sub>3</sub>, SCH<sub>2</sub>CH<sub>3</sub>, SCH<sub>2</sub>CCH, SCH<sub>2</sub>CHCH<sub>2</sub>, SC<sub>3</sub>H<sub>7</sub>,
- 10   NH(CH<sub>3</sub>), N(CH<sub>3</sub>)<sub>2</sub>, NH(CH<sub>2</sub>CH<sub>3</sub>), N(CH<sub>2</sub>CH<sub>3</sub>)<sub>2</sub>, NH(CH<sub>2</sub>CCH),  
   NH(CH<sub>2</sub>CHCH<sub>2</sub>), NH(C<sub>3</sub>H<sub>7</sub>), halogen (F, Cl, Br or I) or X wherein X is  
   -(CH<sub>2</sub>)<sub>m</sub>(O)<sub>n</sub>(CH<sub>2</sub>)<sub>m</sub>N(R<sup>10</sup>)<sub>2</sub> wherein each m is independently 0-2, n is 0-1, and  
   R<sup>10</sup> independently is
- H,
- 15   C<sub>1</sub>-C<sub>15</sub> alkyl, C<sub>2</sub>-C<sub>15</sub> alkenyl, C<sub>6</sub>-C<sub>15</sub> arylalkenyl, C<sub>6</sub>-C<sub>15</sub>  
   arylalkynyl, C<sub>2</sub>-C<sub>15</sub> alkynyl, C<sub>1</sub>-C<sub>6</sub>-alkylamino-C<sub>1</sub>-C<sub>6</sub> alkyl, C<sub>5</sub>-C<sub>15</sub> aralkyl, C<sub>6</sub>-  
   C<sub>15</sub> heteroaralkyl, C<sub>5</sub>-C<sub>6</sub> aryl, C<sub>2</sub>-C<sub>6</sub> heterocycloalkyl,  
   C<sub>2</sub>-C<sub>15</sub> alkyl, C<sub>3</sub>-C<sub>15</sub> alkenyl, C<sub>6</sub>-C<sub>15</sub> arylalkenyl, C<sub>3</sub>-C<sub>15</sub> alkynyl,  
   C<sub>7</sub>-C<sub>15</sub> arylalkynyl, C<sub>1</sub>-C<sub>6</sub>-alkylamino-C<sub>1</sub>-C<sub>6</sub> alkyl, C<sub>5</sub>-C<sub>15</sub> aralkyl, C<sub>6</sub>-C<sub>15</sub>  

20   heteroalkyl or C<sub>3</sub>-C<sub>6</sub> heterocycloalkyl wherein methylene in the alkyl moiety not  
   adjacent to N<sup>6</sup> has been replaced by -O-,  
   optionally both R<sup>10</sup> are joined together with N to form a saturated or  
   unsaturated C<sub>2</sub>-C<sub>5</sub> heterocycle containing one or two N heteroatoms and  
   optionally an additional O or S heteroatom,

25   or one of the foregoing R<sup>10</sup> groups which is substituted with 1 to 3  
   halo, CN or N<sub>3</sub>; but optionally at least one R<sup>10</sup> group is not H;

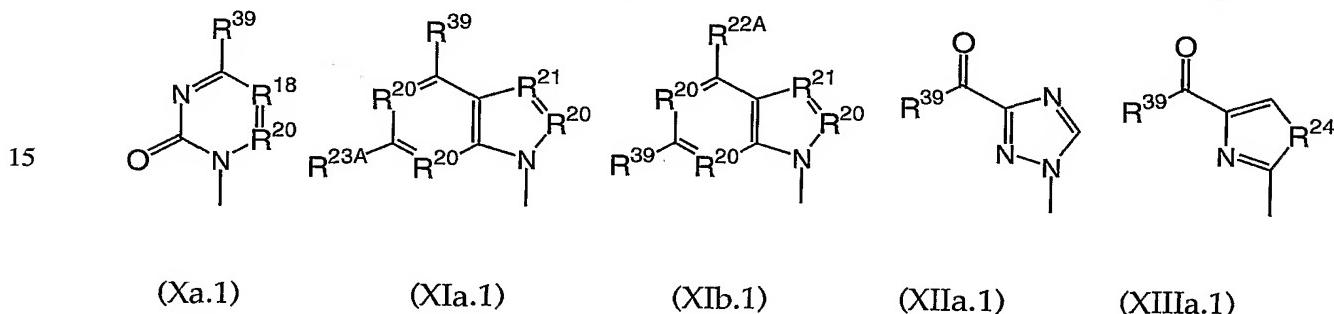
    R<sup>23</sup> is H, OH, F, Cl, Br, I, SCH<sub>3</sub>, SCH<sub>2</sub>CH<sub>3</sub>, SCH<sub>2</sub>CCH, SCH<sub>2</sub>CHCH<sub>2</sub>,  
   SC<sub>3</sub>H<sub>7</sub>, OR<sup>16</sup>, NH<sub>2</sub>, NHR<sup>17</sup> or R<sup>22</sup>; and

    R<sup>24</sup> is O, S or Se.

30   B also includes both protected and unprotected heterocyclic bases,  
   particularly purine and pyrimidine bases. Protecting groups for exocyclic amines  
   and other labile groups are known (Greene et al. "Protective Groups in Organic  
   Synthesis") and include N-benzoyl, isobutyryl, 4,4'-dimethoxytrityl (DMT) and the

5 like. The selection of protecting group will be apparent to the ordinary artisan and will depend upon the nature of the labile group and the chemistry which the protecting group is expected to encounter, e.g. acidic, basic, oxidative, reductive or other conditions. Exemplary protected species are N<sup>4</sup>-benzoylcytosine, N<sup>6</sup>-benzoyladenine, N<sup>2</sup>-isobutyrylguanine and the like.

10 Protected bases have the formulas Xa.1, XIa.1, XIIb.1, XIIa.1 or XIIIa.1



15 20 wherein R<sup>18</sup>, R<sup>20</sup>, R<sup>21</sup>, R<sup>24</sup> have the meanings previously defined; R<sup>22A</sup> is R<sup>39</sup> or R<sup>22</sup> provided that R<sup>22</sup> is not NH<sub>2</sub>; R<sup>23A</sup> is R<sup>39</sup> or R<sup>23</sup> provided that R<sup>23</sup> is not NH<sub>2</sub>; R<sup>39</sup> is NHR<sup>40</sup>, NHC(O)R<sup>36</sup> or CR<sup>41</sup>N(R<sup>38</sup>)<sub>2</sub> wherein R<sup>36</sup> is C<sub>1</sub>-C<sub>19</sub> alkyl, C<sub>1</sub>-C<sub>19</sub> alkenyl, C<sub>3</sub>-C<sub>10</sub> aryl, adamantoyl, alkylanyl, or C<sub>3</sub>-C<sub>10</sub> aryl substituted with 1 or 2 atoms or groups selected from halogen, methyl, ethyl, methoxy, ethoxy, hydroxy and cyano; R<sup>38</sup> is C<sub>1</sub>-C<sub>10</sub> alkyl, or both R<sup>38</sup> together are 1-morpholino, 1-piperidine or 1-pyrrolidine; R<sup>40</sup> is C<sub>1</sub>-C<sub>1a</sub> alkyl, including methyl, ethyl, propyl, isopropyl, butyl, isobutyl, t-butyl, pentyl, hexyl, octyl and decanyl; and R<sup>41</sup> is hydrogen or CH<sub>3</sub>.

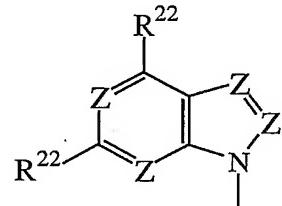
25 For bases of structures XIa.1 and XIIb.1, if R<sup>39</sup> is present at R<sup>22A</sup> or R<sup>23A</sup>, both R<sup>39</sup> groups on the same base will generally be the same. Exemplary R<sup>36</sup> are phenyl, phenyl substituted with one of the foregoing R<sup>36</sup> aryl substituents, -C<sub>10</sub>H<sub>15</sub> (where C<sub>10</sub>H<sub>15</sub> is 2-adamantoyl), -CH<sub>2</sub>-C<sub>6</sub>H<sub>5</sub>, -C<sub>6</sub>H<sub>5</sub>, -CH(CH<sub>3</sub>)<sub>2</sub>, -CH<sub>2</sub>CH<sub>3</sub>, methyl, butyl, t-butyl, heptyl, nonanyl, undecanyl, or undecenyl.

30 Specific bases include hypoxanthine, guanine, adenine, cytosine, inosine, thymine, uracil, xanthine, 8-aza derivatives of 2-aminopurine, 2,6-diaminopurine, 35 2-amino-6-chloropurine, hypoxanthine, inosine and xanthine; 7-deaza-8-aza

5 derivatives of adenine, guanine, 2-aminopurine, 2,6-diaminopurine, 2-amino-6-chloropurine, hypoxanthine, inosine and xanthine; 1-deaza derivatives of 2-aminopurine, 2,6-diaminopurine, 2-amino-6-chloropurine, hypoxanthine, inosine and xanthine; 7-deaza derivatives of 2-aminopurine, 2,6-diaminopurine, 2-amino-6-chloropurine, hypoxanthine, inosine and xanthine; 3-deaza derivatives of 2-aminopurine, 2,6-diaminopurine, 2-amino-6-chloropurine, hypoxanthine, inosine and xanthine; 6-azacytosine; 5-fluorocytosine; 5-chlorocytosine; 5-iodocytosine; 5-bromocytosine; 5-methylcytosine; 5-bromovinyluracil; 5-fluorouracil; 5-chlorouracil; 5-iodouracil; 5-bromouracil; 5-trifluoromethyluracil; 5-methoxymethyluracil; 5-ethynyluracil and 5-propynyluracil.

10 15 Preferably, B is a 9-purinyl residue selected from guanyl, 3-deazaguanyl, 1-deazaguanyl, 8-azaguanyl, 7-deazaguanyl, adenyl, 3-deazaadenyl, 1-dezazaadenyl, 8-azaadenyl, 7-deazaadenyl, 2,6-diaminopurinyl, 2-aminopurinyl, 6-chloro-2-aminopurinyl and 6-thio-2-aminopurinyl, or a B' is a 1-pyrimidinyl residue selected from cytosinyl, 5-halocytosinyl, and 5-(C<sub>1</sub>-C<sub>3</sub>-alkyl)cytosinyl.

20 Preferred B groups have the formula



25

wherein

R<sup>22</sup> independently is halo, oxygen, NH<sub>2</sub>, X or H, but optionally at least one R<sup>22</sup> is X;

X is -(CH<sub>2</sub>)<sub>m</sub>(O)<sub>n</sub>(CH<sub>2</sub>)<sub>m</sub>N(R<sup>10</sup>)<sub>2</sub> wherein m is 0-2, n is 0-1, and

30 R<sup>10</sup> independently is

H,

C<sub>1</sub>-C<sub>15</sub> alkyl, C<sub>2</sub>-C<sub>15</sub> alkenyl, C<sub>6</sub>-C<sub>15</sub> arylalkenyl, C<sub>6</sub>-C<sub>15</sub> arylalkynyl, C<sub>2</sub>-C<sub>15</sub> alkynyl, C<sub>1</sub>-C<sub>6</sub>-alkylamino-C<sub>1</sub>-C<sub>6</sub> alkyl, C<sub>5</sub>-C<sub>15</sub> aralkyl, C<sub>6</sub>-C<sub>15</sub> heteroaralkyl, C<sub>5</sub>-C<sub>6</sub> aryl, C<sub>2</sub>-C<sub>6</sub> heterocycloalkyl,

5           C<sub>2</sub>-C<sub>15</sub> alkyl, C<sub>3</sub>-C<sub>15</sub> alkenyl, C<sub>6</sub>-C<sub>15</sub> arylalkenyl, C<sub>3</sub>-C<sub>15</sub> alkynyl,  
 C<sub>7</sub>-C<sub>15</sub> arylalkynyl, C<sub>1</sub>-C<sub>6</sub>-alkylamino-C<sub>1</sub>-C<sub>6</sub> alkyl, C<sub>5</sub>-C<sub>15</sub> aralkyl, C<sub>6</sub>-C<sub>15</sub>  
 heteroalkyl or C<sub>3</sub>-C<sub>6</sub> heterocycloalkyl wherein methylene in the alkyl moiety not  
 adjacent to N<sup>6</sup> has been replaced by -O-,

10          optionally both R<sup>10</sup> are joined together with N to form a saturated or  
 unsaturated C<sub>2</sub>-C<sub>5</sub> heterocycle containing one or two N heteroatoms and  
 optionally an additional O or S heteroatom,

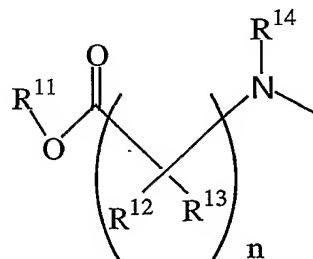
15          or one of the foregoing R<sup>10</sup> groups is substituted with 1 to 3 halo, CN  
 or N<sub>3</sub>; but optionally at least one R<sup>10</sup> group is not H; and

15.        Z is N or CH, provided that the heterocyclic nucleus varies from purine by  
 no more than one Z.

20          E groups represent the aglycons employed in the methoxyphosphonate  
 nucleotide analogues. Preferably, the E group is -CH(CH<sub>3</sub>)CH<sub>2</sub>- or -CH<sub>2</sub>CH<sub>2</sub>- . Also,  
 it is preferred that the side groups at chiral centers in the aglycon be substantially  
 solely in the (R) configuration (except for hydroxymethyl, which is the enriched (S)  
 enantiomer).

25          R<sup>1</sup> is an *in vivo* hydrolyzable oxyester having the structure -OR<sup>35</sup> or -OR<sup>6</sup>  
 wherein R<sup>35</sup> is defined in column 64, line 49 of U.S. Patent No. 5,798,340, herein  
 incorporated by reference, and R<sup>6</sup> is defined above. Preferably R<sup>1</sup> is aryloxy,  
 ordinarily unsubstituted or para-substituted (as defined in R<sup>6</sup>) phenoxy.

30          R<sup>2</sup> is an amino acid residue, optionally provided that any carboxy group  
 linked by less than about 5 atoms to the amide N is esterified. R<sup>2</sup> typically has the  
 structure



(8)

35        wherein

5 n is 1 or 2;

R<sup>11</sup> is R<sup>6</sup> or H; preferably R<sup>6</sup> = C<sub>1</sub>-C<sub>9</sub> alkyl; C<sub>3</sub>-C<sub>9</sub> alkyl substituted independently with OH, halogen, O or N; C<sub>3</sub>-C<sub>6</sub> aryl; C<sub>3</sub>-C<sub>6</sub> aryl which is independently substituted with OH, halogen, O or N; or C<sub>3</sub>-C<sub>6</sub> arylalkyl which is independently substituted with OH, halogen, O or N;

10 R<sup>12</sup> independently is H or C<sub>1</sub>-C<sub>9</sub> alkyl which is unsubstituted or substituted by substituents independently selected from the group consisting of OH, O, N, COOR<sup>11</sup> and halogen; C<sub>3</sub>-C<sub>6</sub> aryl which is unsubstituted or substituted by substituents independently selected from the group consisting of OH, O, N, COOR<sup>11</sup> and halogen; or C<sub>3</sub>-C<sub>6</sub> aryl-alkyl which is unsubstituted or substituted by substituents independently selected from the group consisting of OH, O, N, COOR<sup>11</sup> and halogen;

15 R<sup>13</sup> independently is C(O)-OR<sup>11</sup>; amino; amide; guanidinyl; imidazolyl; indolyl; sulfoxide; phosphoryl; C<sub>1</sub>-C<sub>3</sub> alkylamino; C<sub>1</sub>-C<sub>3</sub> alkyldiamino; C<sub>1</sub>-C<sub>6</sub> alkenylamino; hydroxy; thiol; C<sub>1</sub>-C<sub>3</sub> alkoxy; C<sub>1</sub>-C<sub>3</sub> alkthiol; (CH<sub>2</sub>)<sub>n</sub>COOR<sup>11</sup>; C<sub>1</sub>-C<sub>6</sub> 20 alkyl which is unsubstituted or substituted with OH, halogen, SH, NH<sub>2</sub>, phenyl, hydroxyphenyl or C<sub>7</sub>-C<sub>10</sub> alkoxyphenyl; C<sub>2</sub>-C<sub>6</sub> alkenyl which is unsubstituted or substituted with OH, halogen, SH, NH<sub>2</sub>, phenyl, hydroxyphenyl or C<sub>7</sub>-C<sub>10</sub> alkoxyphenyl; and C<sub>6</sub>-C<sub>12</sub> aryl which is unsubstituted or substituted with OH, halogen, SH, NH<sub>2</sub>, phenyl, hydroxyphenyl or C<sub>7</sub>-C<sub>10</sub> alkoxyphenyl; and

25 R<sup>14</sup> is H or C<sub>1</sub>-C<sub>9</sub> alkyl or C<sub>1</sub>-C<sub>9</sub> alkyl independently substituted with OH, halogen, COOR<sup>11</sup>, O or N; C<sub>3</sub>-C<sub>6</sub> aryl; C<sub>3</sub>-C<sub>6</sub> aryl which is independently substituted with OH, halogen, COOR<sup>11</sup>, O or N; or C<sub>3</sub>-C<sub>6</sub> arylalkyl which is independently substituted with OH, halogen, COOR<sup>11</sup>, O or N.

30 Preferably, R<sup>11</sup> is C<sub>1</sub>-C<sub>6</sub> alkyl, most preferably isopropyl, R<sup>13</sup> is the side chain of a naturally occurring amino acid, n = 1, R<sup>12</sup> is H and R<sup>14</sup> is H. In the compound of structure (2), the invention includes metabolites in which the phenoxy and isopropyl esters have been hydrolyzed to -OH. Similarly, the de-esterified enriched phosphonoamide metabolites of compounds (5a), 5(b) and (6) are included within the scope of this invention.

35 Aryl and "O" or "N" substitution are defined in column 16, lines 42-58, of

5 United States Patent No. 5,798,340.

Typically, the amino acids are in the natural or *l* amino acids. Suitable specific examples are set forth in U. S. Patent No. 5,798,340, for instance Table 4 and col. 8-10 therein.

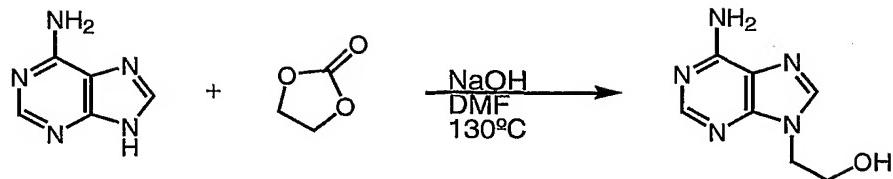
Alkyl as used herein, unless stated to the contrary, is a normal, secondary, 10 tertiary or cyclic hydrocarbon. Unless stated to the contrary alkyl is C<sub>1</sub>-C<sub>12</sub>. Examples are -CH<sub>3</sub>, -CH<sub>2</sub>CH<sub>3</sub>, -CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>, -CH(CH<sub>3</sub>)<sub>2</sub>, -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), -CH<sub>2</sub>CH(CH<sub>3</sub>)<sub>2</sub>, -CH(CH<sub>3</sub>)CH<sub>2</sub>CH<sub>3</sub>, -C(CH<sub>3</sub>)<sub>3</sub>, -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>, -CH(CH<sub>3</sub>)CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>, -CH(CH<sub>2</sub>CH<sub>3</sub>)<sub>2</sub>, -C(CH<sub>3</sub>)<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), -CH(CH<sub>3</sub>)CH(CH<sub>3</sub>)<sub>2</sub>, -CH<sub>2</sub>CH<sub>2</sub>CH(CH<sub>3</sub>)<sub>2</sub>, -CH<sub>2</sub>CH(CH<sub>3</sub>)CH<sub>2</sub>CH<sub>3</sub>, 15 -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>, -CH(CH<sub>3</sub>)CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>, -CH(CH<sub>2</sub>CH<sub>3</sub>)(CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), -C(CH<sub>3</sub>)<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>, -CH(CH<sub>3</sub>)CH(CH<sub>3</sub>)CH<sub>2</sub>CH<sub>3</sub>, -CH(CH<sub>3</sub>)CH<sub>2</sub>CH(CH<sub>3</sub>)<sub>2</sub>, -C(CH<sub>3</sub>)(CH<sub>2</sub>CH<sub>3</sub>)<sub>2</sub>, -CH(CH<sub>2</sub>CH<sub>3</sub>)CH(CH<sub>3</sub>)<sub>2</sub>, -C(CH<sub>3</sub>)<sub>2</sub>CH(CH<sub>3</sub>)<sub>2</sub>, and -CH(CH<sub>3</sub>)C(CH<sub>3</sub>)<sub>3</sub>. Alkenyl and alkynyl are defined in the same fashion, but 20 contain at least one double or triple bond, respectively.

Where enol or keto groups are disclosed, the corresponding tautomers are to be construed as taught as well.

The prodrug compounds of this invention are provided in the form of free base or the various salts enumerated in U. S. Patent No. 5,798,340, and are 25 formulated with pharmaceutically acceptable excipients or solvating diluents for use as pharmaceutical products also as set forth in U. S. Patent No. 5,798,340. These prodrugs have the antiviral and utilities already established for the parent drugs (see U. S. Patent 5,798,340 and other citations relating to the methoxyphosphonate nucleotide analogues). It will be understood that the diastereomer of structure (4) 30 at least is useful as an intermediate in the chemical production of the parent drug by hydrolysis *in vitro*, regardless of its relatively unselective character as revealed in the studies herein.

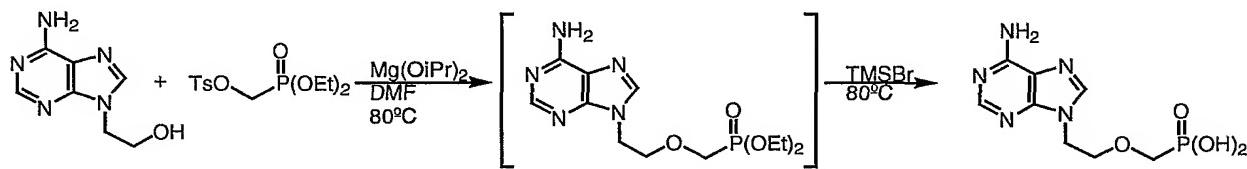
The invention will be more fully understood by reference to the following examples:

5

Example 1a

**Adenine to PMEA using Magnesium Isopropoxide.** To a suspension of adenine (16.8g, 0.124 mol) in DMF (41.9 ml) was added ethylene carbonate (12.1g, 0.137 mol) and sodium hydroxide (.100g, 0.0025 mol). The mixture was heated at 130°C overnight. The reaction was cooled to below 50°C and toluene (62.1 ml) was added. The slurry was further cooled to 5°C for 2 hours, filtered, and rinsed with toluene (2x). The wet solid was dried *in vacuo* at 65°C to yield 20.0g (90%) of 9-(2-hydroxyethyl)adenine as an off-white solid. Mp: 238-240°C.

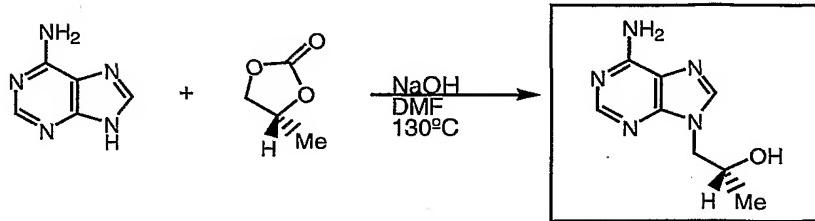
15



9-(2-Hydroxyethyl)adenine (HEA) (20.0g, 0.112 mol) was suspended in DMF (125 ml) and heated to 80°C. Magnesium isopropoxide (11.2g, 0.0784 mol), or alternatively magnesium *t*-butoxide, was added to the mixture followed by diethyl *p*-toluenesulfonyloxyethylphosphonate (66.0g, 0.162 mol) over one hour. The mixture was stirred at 80°C for 7 hours. 30 ml of volatiles were removed via vacuum distillation and the reaction was recharged with 30 ml of fresh DMF. After cooling to room temperature, bromotrimethylsilane (69.6g, 0.450 mol) was added and the mixture heated to 80°C for 6 hours. The reaction was concentrated to yield a thick gum. The gum was dissolved into 360 ml water, extracted with 120 ml dichloromethane, adjusted to pH 3.2 with sodium hydroxide, and the resulting slurry stirred at room temperature overnight. The slurry was cooled to 4°C for one hour. The solids were isolated by filtration, washed with water (2x), and dried *in*

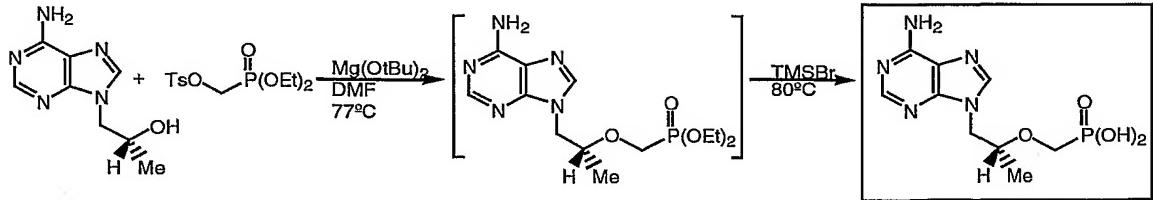
5 *vacuo* at 56°C to yield 20g (65.4%) of 9-[2-(phosphonomethoxy)ethyl]adenine (PMEA) as a white solid. Mp: > 200°C dec.  $^1\text{H}$  NMR ( $\text{D}_2\text{O}$ ) • 3.49 (t, 2H); 3.94 (t, 2H); 4.39 (t, 2H); 8.13 (s, 1H); 8.22 (s, 1H).

Example 1b



10

**Adenine to PMPA using Magnesium t-Butoxide.** To a suspension of adenine (40g, 0.296 mol) in DMF (41.9 ml) was added (*R*)-propylene carbonate (34.5g, 0.338 mol) and sodium hydroxide (.480g, 0.012 mol). The mixture was heated at 130°C overnight. The reaction was cooled to 100°C and toluene (138 ml) was added followed by methanesulfonic acid (4.7g, 0.049 mol) while maintaining the reaction temperature between 100-110°C. Additional toluene (114 ml) was added to create a homogeneous solution. The solution was cooled to 3°C over 7 hours and then held at 3°C for one hour. The resulting solid was isolated by filtration and rinsed with acetone (2x). The wet solid was dried *in vacuo* at 80°C to yield 42.6g (75%) of (*R*)-9-[2-(hydroxy)propyl]adenine (HPA) as an off-white solid. Mp: 188-190°C.



(*R*)-9-[2-(hydroxy)propyl]adenine (HPA) (20.0g, 0.104 mol) was suspended in DMF (44.5 ml) and heated to 65°C. Magnesium t-butoxide (14.2g, 0.083 mol), or

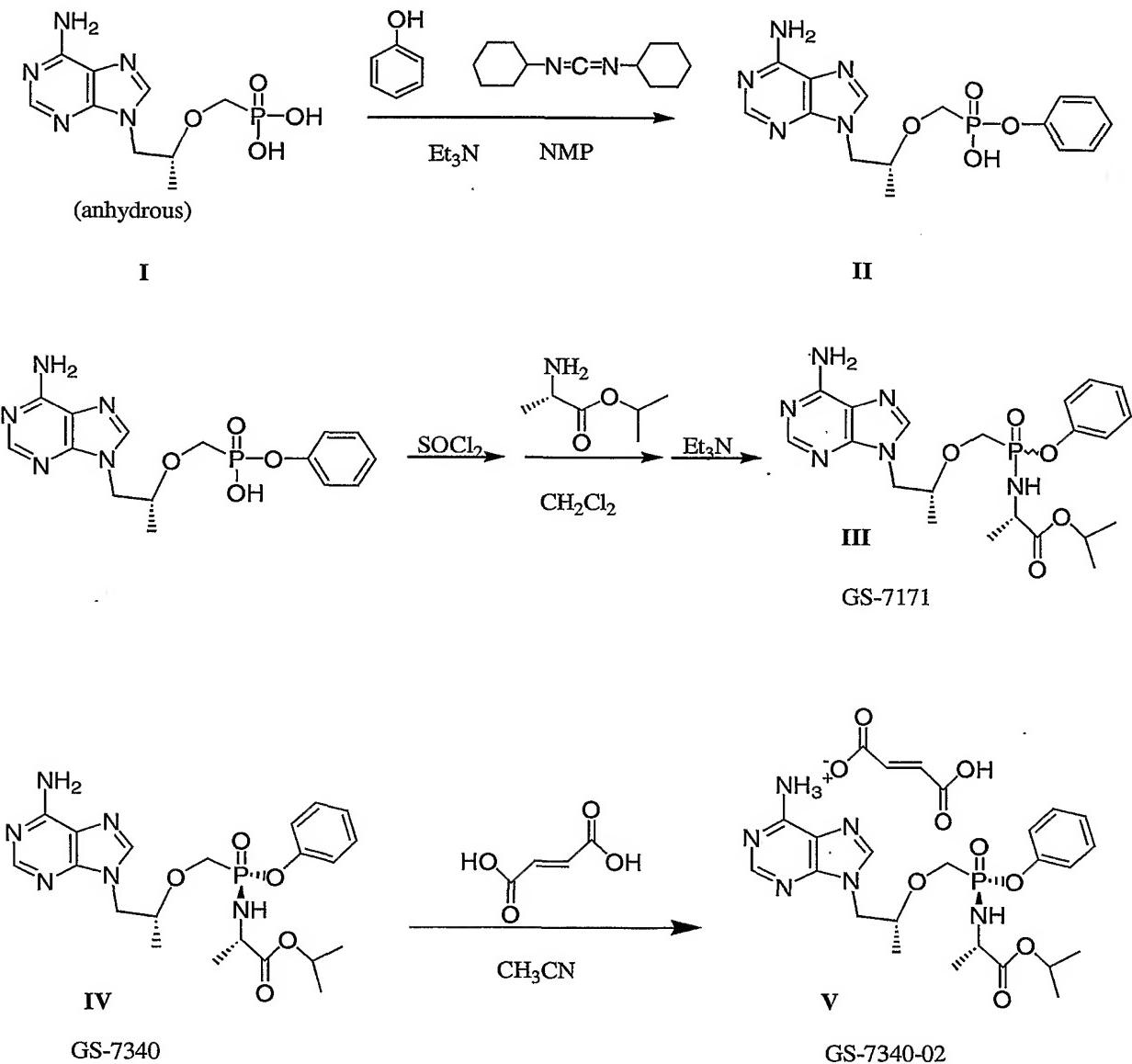
25 alternatively magnesium isopropoxide, was added to the mixture over one hour followed by diethyl *p*-toluenesulfonyloxymethylphosphonate (66.0g, 0.205 mol)

5 over two hours while the temperature was kept at 78°C. The mixture was stirred at  
75°C for 4 hours. After cooling to below 50°C, bromotrimethylsilane (73.9g, 0.478  
mol) was added and the mixture heated to 77°C for 3 hours. When complete, the  
reaction was heated to 80°C and volatiles were removed via atmospheric  
distillation. The residue was dissolved into water (120 ml) at 50°C and then  
10 extracted with ethyl acetate (101 ml). The pH of the aqueous phase was adjusted to  
pH 1.1 with sodium hydroxide, seeded with authentic (R)-PMPA, and the pH of  
the aqueous layer was readjusted to pH 2.1 with sodium hydroxide. The resulting  
slurry was stirred at room temperature overnight. The slurry was cooled to 4°C for  
three hours. The solid was isolated by filtration, washed with water (60 ml), and  
15 dried *in vacuo* at 50°C to yield 18.9g (63.5%) of crude(R)-9-[2-  
(phosphonomethoxy)propyl]adenine (PMPA) as an off-white solid.

The crude(R)-9-[2-(phosphonomethoxy)propyl]adenine was heated at reflux in  
water (255 ml) until all solids dissolved. The solution was cooled to room  
20 temperature over 4 hours. The resulting slurry was cooled at 4°C for three hours.  
The solid was isolated by filtration, washed with water (56 ml) and acetone (56 ml),  
and dried *in vacuo* at 50°C to yield 15.0g (50.4%) of (R)-9-[2-  
(phosphonomethoxy)propyl]adenine (PMPA) as a white solid. Mp: 278-280°C.

Example 2  
Preparation of GS-7171 (III)

Scheme 1



5 A glass-lined reactor was charged with anhydrous PMPA, (I) (14.6 kg, 50.8 mol), phenol (9.6 kg, 102 mol), and 1-methyl-2-pyrrolidinone (39 kg). The mixture was heated to 85°C and triethylamine (6.3 kg, 62.3 mol) added. A solution of 1,3-dicyclohexylcarbodiimide (17.1 kg, 82.9 mol) in 1-methyl-2-pyrrolidinone (1.6 kg) was then added over 6 hours at 100°C. Heating was continued for 16 hours. The  
10 reaction was cooled to 45°C, water (29 kg) added, and cooled to 25°C. Solids were removed from the reaction by filtration and rinsed with water (15.3 kg). The combined filtrate and rinse was concentrated to a tan slurry under reduced pressure, water (24.6 kg) added, and adjusted to pH = 11 with NaOH (25% in water). Fines were removed by filtration through diatomaceous earth (2 kg)  
15 followed by a water (4.4 kg) rinse. The combined filtrate and rinse was extracted with ethyl acetate (28 kg). The aqueous solution was adjusted to pH = 3.1 with HCl (37% in water) (4 kg). Crude II was isolated by filtration and washed with methanol (12.7 kg). The crude II wet cake was slurried in methanol (58 kg). Solids were isolated by filtration, washed with methanol (8.5 kg), and dried under  
20 reduced pressure to yield 9.33 kg II as a white powder:  $^1\text{H}$  NMR ( $\text{D}_2\text{O}$ )  $\delta$  1.2 (d, 3H), 3.45 (q, 2H), 3.7 (q, 2H), 4 (m, 2H), 4.2 (q, 2H), 4.35 (dd, 2H), 6.6 (d, 2H), 7 (t, 1H), 7.15 (t, 2H), 8.15 (s, 1H), 8.2 (s, 1H);  $^{31}\text{P}$  NMR ( $\text{D}_2\text{O}$ )  $\delta$  15.0 (decoupled).

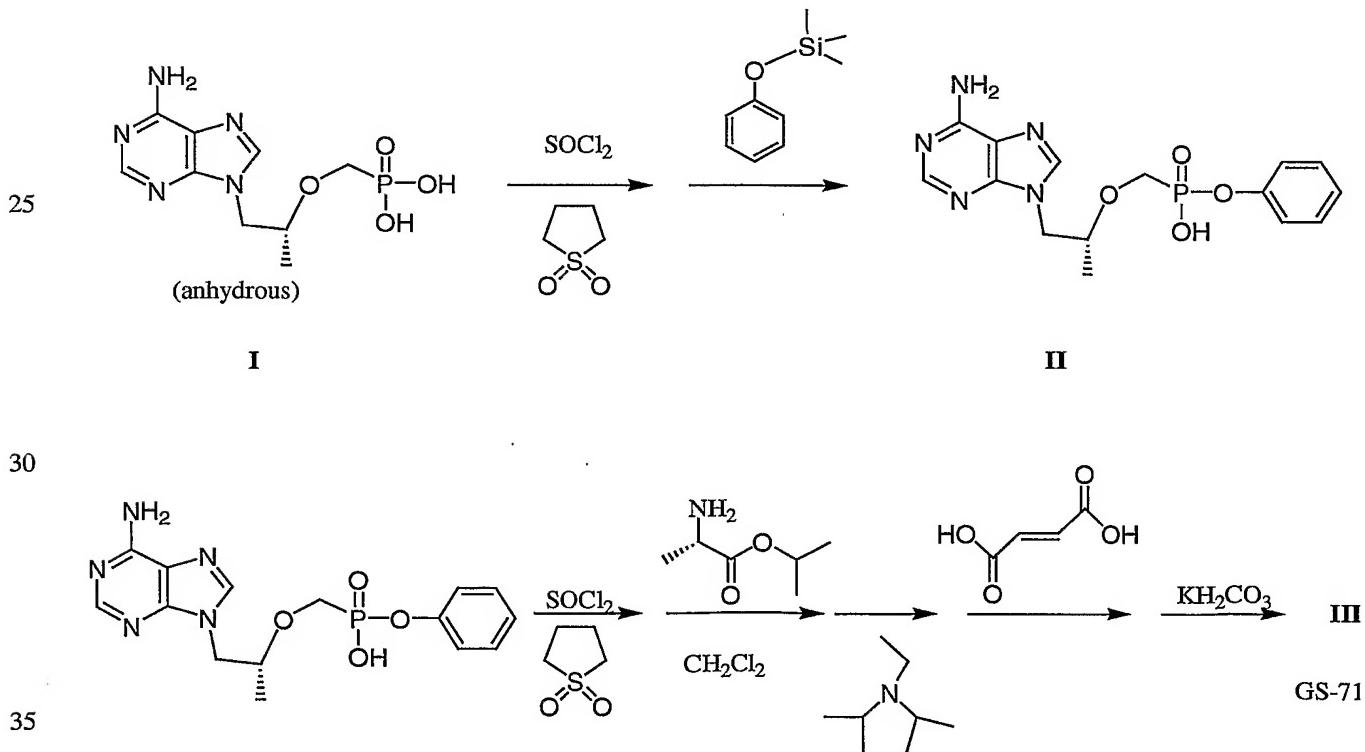
GS-7171 (III). (Scheme 1) A glass-lined reactor was charged with monophenyl  
25 PMPA, (II), (9.12 kg, 25.1 mol) and acetonitrile (30.7 kg). Thionyl chloride (6.57 kg, 56.7 mol) was added below 50°C. The mixture was heated at 75°C until solids dissolved. Reaction temperature was increased to 80°C and volatiles (11.4 kg) collected by atmospheric distillation under nitrogen. The pot residue was cooled to 25°C, dichloromethane (41 kg) added, and cooled to -29°C. A solution of (L)-  
30 alanine isopropyl ester (7.1 kg, 54.4 mol) in dichloromethane (36 kg) was added over 60 minutes at -18°C followed by triethylamine (7.66 kg, 75.7 mol) over 30 minutes at -18 to -11°C. The reaction mixture was warmed to room temperature and washed five times with sodium dihydrogenphosphate solution (10% in water, 15.7 kg each wash). The organic solution was dried with anhydrous sodium sulfate  
35 (18.2 kg), filtered, rinsed with dichloromethane (28 kg), and concentrated to an oil

5 under reduced pressure. Acetone (20 kg) was charged to the oil and the mixture concentrated under reduced pressure. Acetone (18.8 kg) was charged to the resulting oil. Half the product solution was purified by chromatography over a 38 x 38 cm bed of 22 kg silica gel 60, 230 to 400 mesh. The column was eluted with 480 kg acetone. The purification was repeated on the second half of the oil using fresh  
10 silica gel and acetone. Clean product bearing fractions were concentrated under reduced pressure to an oil. Acetonitrile (19.6 kg) was charged to the oil and the mixture concentrated under reduced pressure. Acetonitrile (66.4 kg) was charged and the solution chilled to 0 to -5°C for 16 hours. Solids were removed by filtration and the filtrate concentrated under reduced pressure to 5.6 kg III as a dark oil: <sup>1</sup>H  
15 NMR ( $\text{CDCl}_3$ ) δ 1.1 (m, 12H), 3.7 (m, 1H), 4.0 (m, 5H), 4.2 (m, 1H), 5.0 (m, 1H), 6.2 (s, 2H), 7.05 (m, 5H), 8.0 (s, 1H), 8.25 (d, 1H); <sup>31</sup>P NMR ( $\text{CDCl}_3$ ) δ 21.0, 22.5 (decoupled).

### Alternate Method for GS-7171(III)

20

**Scheme 2**



5   **Monophenyl PMPA (II).** A round-bottom flask with reflux condenser and  
nitrogen inlet was placed in a 70°C oil bath. The flask was charged with anhydrous  
PMPA (I) (19.2 g, 67 mmol), *N,N*-dimethylformamide (0.29 g, 3.3 mmol), and  
tetramethylene sulfone (40 mL). Thionyl chloride (14.2 g, 119 mmol) was added  
over 4 hours. Heating was increased to 100°C over the same time. A homogeneous  
10   solution resulted. Phenoxytrimethylsilane (11.7 g, 70 mmol) was added to the  
solution over 5 minutes. Heating in the 100°C oil bath continued for two hours  
more. The reaction was poured into rapidly stirring acetone (400 mL) with cooling  
at 0°C. Solids were isolated by filtration, dried under reduced pressure, and  
dissolved in methanol (75 mL). The solution pH was adjusted to 3.0 with  
15   potassium hydroxide solution (45% aq.) with cooling in ice/water. The resulting  
solids were isolated by filtration, rinsed with methanol, and dried under reduced  
pressure to 20.4 g II (Scheme 2) as a white powder.

20   **GS-7171 (III).** Monophenyl PMPA (II) (3 g, 8.3 mmol), tetramethylene sulfone (5  
mL), and *N,N*-dimethylformamide (1 drop) were combined in a round bottom flask  
in a 40°C oil bath. Thionyl chloride (1.96 g, 16.5 mmol) was added. After 20  
minutes the clear solution was removed from heat, diluted with dichloromethane  
(10 ml), and added to a solution of (L)-alanine isopropyl ester (5g, 33 mmol) and  
diisopropylethylamine (5.33 g, 41 mmol) in dichloromethane (20 mL) at -10°C. The  
25   reaction mixture was warmed to room temperature and washed three times with  
sodium dihydrogenphosphate solution (10% aq., 10 mL each wash). The organic  
solution was dried over anhydrous sodium sulfate and concentrated under  
reduced pressure to a oil. The oil was combined with fumaric acid (0.77g, 6.6  
mmol) and acetonitrile (40 mL) and heated to reflux to give a homogeneous  
30   solution. The solution was cooled in an ice bath and solids isolated by filtration.  
The solid GS-7171 fumarate salt was dried under reduced pressure to 3.7 g. The  
salt (3.16 g, 5.3 mmol) was suspended in dichloromethane (30 mL) and stirred with  
potassium carbonate solution (5 mL, 2.5 M in water) until the solid dissolved. The  
organic layer was isolated, then washed with water (5 mL), dried over anhydrous

5 sodium sulfate, and concentrated under reduced pressure to afford 2.4 g III as a tan foam.

### Example 3

#### A. Diastereomer Separation by Batch Elution Chromatography

10

The diastereomers of GS-7171 (III) were resolved by batch elution chromatography using a commercially available Chiralpak AS, 20 µm, 21 x 250 mm semi-preparative HPLC column with a Chiralpak AS, 20 µm, 21 x 50 mm guard column. Chiralpak® AS is a proprietary packing material manufactured by Diacel 15 and sold in North America by Chiral Technologies, Inc. (U. S. Patent Nos. 5,202,433, RE 35,919, 5,434,298, 5,434,299 and 5,498,752). Chiralpak AS is a chiral stationary phase (CSP) comprised of amylosetris[(S)- $\alpha$ -methylbenzyl carbamate] coated onto a silica gel support.

20 The GS-7171 diastereomeric mixture was dissolved in mobile phase, and approximately 1 g aliquots of GS-7171 were pumped onto the chromatographic system. The undesired diastereomer, designated GS-7339, was the first major broad (approx. 15 min. duration) peak to elute from the column. When the GS-7339 peak had finished eluting, the mobile phase was immediately switched to 100% 25 methyl alcohol, which caused the desired diastereomer, designated GS-7340 (IV), to elute as a sharp peak from the column with the methyl alcohol solvent front. The methyl alcohol was used to reduce the over-all cycle time. After the first couple of injections, both diastereomers were collected as a single large fractions containing one of the purified diastereomers (>99.0% single diastereomer). The mobile phase solvents were removed *in vacuo* to yield the purified diastereomer as a friable foam.

30 About 95% of the starting GS-7171 mass was recovered in the two diastereomer fractions. The GS-7340 fraction comprised about 50% of the total recovered mass.

5 The chromatographic conditions were as follows:

Mobile Phase(Initial)	:	GS-7171 – Acetonitrile : Isopropyl Alcohol (90:10)
(Final)	:	100% Methyl Alcohol
Flow	:	10 mL/minute
10 Run Time	:	About 45 minute
Detection	:	UV at 275 nm
Temperature	:	Ambient
Elution Profile	:	GS-7339 (diastereomer B) GS-7340 (diastereomer A; (IV))

15

#### B. Diastereomer Separation of GS-7171 by SMB Chromatography

For a general description of simulated moving bed (SMB) chromatography, see Strube et al., "Organic Process Research and Development" 2:305-319 (1998).

20

**GS-7340 (IV).** GS-7171 (III), 2.8 kg, was purified by simulated moving bed chromatography over 10 cm by 5 cm beds of packing (Chiral Technologies Inc., 20 micron Chiraldak AS coated on silica gel) (1.2 kg). The columns were eluted with 30% methanol in acetonitrile. Product bearing fractions were concentrated to a solution of IV in acetonitrile (2.48 kg). The solution solidified to a crystalline mass wet with acetonitrile on standing. The crystalline mass was dried under reduced pressure to a tan crystalline powder, 1.301 kg IV, 98.7% diastereomeric purity: mp 117 – 120°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.15 (m 12H), 3.7 (t, 1H), 4.0 (m, 5H), 4.2 (dd, 1H), 5.0 (m, 1H), 6.05 (s, 2H), 7.1 (m, 5H), 8.0 (s, 1H), 8.2 (s, 1H); <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ 21.0 (decoupled).

25

30

#### C. Diastereomer Separation by C18 RP-HPLC

GS-7171 (III) was chromatographed by reverse phase HPLC to separate the diastereomers using the following summary protocol.

5

Chromatographic column: Phenomenex Luna™ C18(2), 5 µm, 100 Å pore size, (Phenomenex, Torrance, CA), or equivalent

Guard column: Pellicular C18 (Alltech, Deerfield, IL), or equivalent

10 Mobile Phase: A — 0.02% (85%) H<sub>3</sub>PO<sub>4</sub> in water : acetonitrile (95:5)  
B — 0.02% (85%) H<sub>3</sub>PO<sub>4</sub> in water : acetonitrile (50:50)

15 Mobile Phase Gradient:

Time	% Mobile Phase A	% Mobile Phase B
0	100	0
5	100	0
7	70	30
32	70	30
40	0	100
50	0	100

Run Time: 50 minutes

Equilibration Delay: 10 min at 100% mobile phase A

20 Flow Rate: 1.2 mL/min

Temperature: Ambient

Detection: UV at 260 nm

Sample Solution: 20 mM sodium phosphate buffer, pH 6

Retention Times: GS-7339, about 25 minutes

25 GS-7340, about 27 minutes

#### D. Diastereomer Separation by Crystallization

30 **GS-7340 (IV).** A solution of GS-7171 (III) in acetonitrile was concentrated to an amber foam (14.9g) under reduced pressure. The foam was dissolved in acetonitrile (20 mL) and seeded with a crystal of IV. The mixture was stirred overnight, cooled to 5°C, and solids isolated by filtration. The solids were dried to 2.3 g IV as white crystals, 98% diastereomeric purity (<sup>31</sup>P NMR): <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.15 (m 12H), 3.7 (t, 1H), 3.95 (m, 2H), 4.05 (m, 2H), 4.2 (m, 2H), 5.0 (m, 1H), 6.4 (s, 2H), 7.1 (m, 5H), 8.0 (s, 1H), 8.2 (s, 1H); <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ 19.5 (decoupled). X-ray crystal analysis of a single crystal selected from this product yielded the following data:

5

	Crystal Color, Habit	colorless, column
	Crystal Dimensions	0.25 X 0.12 X 0.08 mm
	Crystal System	orthorhombic
	Lattice Type	Primitive
10	Lattice Parameters	$a = 8.352(1) \text{ \AA}$ $b = 15.574(2) \text{ \AA}$ $c = 18.253(2) \text{ \AA}$ $V = 2374.2(5) \text{ \AA}^3$
15	Space Group	P2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub> (#19)
	Z value	4
	D <sub>calc</sub>	1.333 g/cm <sup>3</sup>
	F <sub>000</sub>	1008.00
	$\mu(\text{MoK}\alpha)$	1.60 cm <sup>-1</sup>

20

Example 4

## Preparation of Fumarate Salt of GS-7340

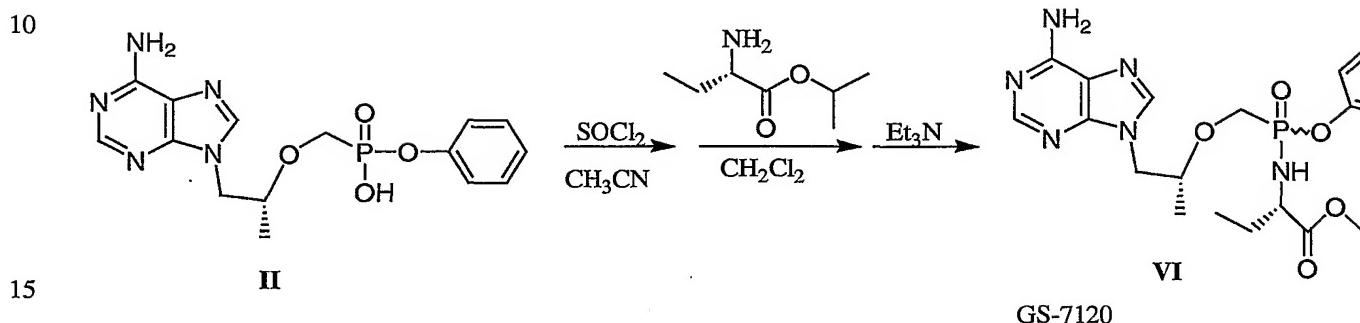
GS-7340-02 (V). (Scheme 1) A glass-lined reactor was charged with GS-7340 (IV), (1.294 kg, 2.71 mol), fumaric acid (284 g, 2.44 mol), and acetonitrile (24.6 kg). The mixture was heated to reflux to dissolve the solids, filtered while hot and cooled to 5°C for 16 hours. The product was isolated by filtration, rinsed with acetonitrile (9.2 kg), and dried to 1329 g (V) as a white powder: mp 119.7 - 121.1°C;  $[\alpha]_D^{20} -41.7^\circ$  (c 1.0, acetic acid).

5

### Example 5

## Preparation of GS-7120 (VI)

### Scheme 3



A 5 L round bottom flask was charged with monophenyl PMPA, (II), (200 g, 0.55 mol) and acetonitrile (0.629 kg). Thionyl chloride (0.144 kg, 1.21 mol) was added below 27°C. The mixture was heated at 70°C until solids dissolved. Volatiles (0.45 L) were removed by atmospheric distillation under nitrogen. The pot residue was cooled to 25°C, dichloromethane (1.6 kg) was added and the mixture was cooled to -20°C. A solution of (L)- $\alpha$  aminobutyric acid ethyl ester (0.144 kg, 1.1 mol) in dichloromethane (1.33 kg) was added over 18 minutes at -20 to -10°C followed by triethylamine (0.17 kg, 1.65 mol) over 15 minutes at -8 to -15°C. The reaction mixture was warmed to room temperature and washed four times with sodium dihydrogenphosphate solution (10% aq., 0.3 L each wash). The organic solution was dried with anhydrous sodium sulfate (0.5 kg) and filtered. The solids were rinsed with dichloromethane (0.6 kg) and the combined filtrate and rinse was concentrated to an oil under reduced pressure. The oil was purified by chromatography over a 15 x 13 cm bed of 1.2 kg silica gel 60, 230 to 400 mesh. The column was eluted with a gradient of dichloromethane and methanol. Product bearing fractions were concentrated under reduced pressure to afford 211 g VI (Scheme 3) as a tan foam.

5

Example 5a

## Diastereomer Separation of GS-7120 by Batch Elution Chromatography

The diastereomeric mixture was purified using the conditions described for GS-7171 in Example 3A except for the following:

10

Mobile Phase      (Initial) : GS-7120 – Acetonitrile : Isopropyl Alcohol (98:2)  
                      (Final) : 100% Methyl Alcohol  
Elution Profile    : GS-7341 (diastereomer B)  
                      : GS-7342 (diastereomer A)

15

Example 6

## Diastereomer Separation of GS-7120 by Crystallization

20

A 1 L round bottom flask was charged with monophenyl PMPA, (II), (50 g, 0.137 mol) and acetonitrile (0.2 L). Thionyl chloride (0.036 kg, 0.303 mol) was added with a 10°C exotherm. The mixture was heated to reflux until solids dissolved. Volatiles (0.1 L) were removed by atmospheric distillation under nitrogen. The pot residue was cooled to 25°C, dichloromethane (0.2 kg) was added, and the mixture was cooled to -20°C. A solution of (L)- $\alpha$  aminobutyric acid ethyl ester (0.036 kg, 0.275 mol) in dichloromethane (0.67 kg) was added over 30 minutes at -20 to -8°C followed by triethylamine (0.042 kg, 0.41 mol) over 10 minutes at up to -6°C. The reaction mixture was warmed to room temperature and washed four times with sodium dihydrogenphosphate solution (10% aq., 0.075 L each wash). The organic solution was dried with anhydrous sodium sulfate (0.1 kg) and filtered. The solids were rinsed with ethyl acetate (0.25 L, and the combined filtrate and rinse was concentrated to an oil under reduced pressure. The oil was diluted with ethyl acetate (0.25 L), seeded, stirred overnight, and chilled to -15°C. The solids were isolated by filtration and dried under reduced pressure to afford 17.7 g of GS-7342 (Table 5) as a tan powder:  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.95 (t, 3H), 1.3 (m, 6H), 1.7, (m, 2H),

5      3.7 (m, 2H), 4.1(m, 6H), 4.4 (dd, 1H), 5.8 (s, 2H), 7.1 (m, 5H), 8.0 (s, 1H), 8.4 (s, 1H);  
31P NMR (CDCl<sub>3</sub>) δ 21 (decoupled).

Example 7

Diastereomer Separation of GS-7097

10

The diastereomeric mixture was purified using the conditions described for GS-7171 (Example 3A) except for the following:

15      Mobile Phase                    (Initial) : GS-7120 – Acetonitrile : Isopropyl Alcohol (95:5)  
    (Final) : 100% Methyl Alcohol  
Elution Profile                            : GS-7115 (diastereomer B)  
    : GS-7114 (diastereomer A)

20

Example 8

Alternative Procedure for Preparation of GS-7097

GS-7097: Phenyl PMPA, Ethyl L-Alanyl Amide. Phenyl PMPA (15.0 g, 41.3 mmol), L-alanine ethyl ester hydrochloride (12.6 g, 83 mmol) and triethylamine (11.5 mL, 83 mmol) were slurried together in 500 mL pyridine under dry N<sub>2</sub>. This suspension was combined with a solution of triphenylphosphine (37.9 g, 145 mmol), Aldrithiol 2 (2,2'-dipyridyl disulfide) (31.8 g, 145 mmol), and 120 mL pyridine. The mixture was heated at an internal temperature of 57°C for 15 hours. The complete reaction was concentrated under vacuum to a yellow paste, 100 g.

30      The paste was purified by column chromatography over a 25 x 11 cm bed of 1.1 kg silica gel 60, 230 to 400 mesh. The column was eluted with 8 liters of 2% methanol in dichloromethane followed by a linear gradient over a course of 26 liters eluent up to a final composition of 13% methanol. Clean product bearing fractions were concentrated to yield 12.4 g crude (5), 65% theory. This material was contaminated with about 15% (weight) triethylamine hydrochloride by <sup>1</sup>H NMR. The contamination was removed by dissolving the product in 350 mL ethyl acetate, extracting with 20 mL water, drying the organic solution over anhydrous sodium

35

5 sulfate, and concentrating to yield 11.1 g pure GS-7097 as a white solid, 58% yield. The process also is employed to synthesize the diastereomeric mixture of GS-7003a and GS-7003b (the phenylalanyl amide) and the mixture GS-7119 and GS-7335 (the glycyl amide). These diastereomers are separated using a batch elution procedure such as shown in Example 3A, 6 and 7.

10

### Example 9

#### *In Vitro* Studies of Prodrug Diastereomers

15 The *in vitro* anti-HIV-1 activity and cytotoxicity in MT-2 cells and stability in human plasma and MT-2 cell extracts of GS-7340 (freebase) and tenofovir disoproxil fumarate (TDF), are shown in Table 1. GS-7340 shows a 10-fold increase in antiviral activity relative to TDF and a 200-fold increase in plasma stability. This greater plasma stability is expected to result in higher circulating levels of GS-7340 than TDF after oral administration.

20

**Table 1. *In Vitro* Activity and Stability**

	<b>HIV-1 Activity</b>	<b>Cytotoxicity</b>	<b>Stability T 1/2 (min)</b>			
			<b>IC<sub>50</sub>μM</b>	<b>Human Plasma</b>	<b>MT-2 Cell Extract</b>	<b>(P/MT-2)</b>
GS 7340	0.005	> 40	90.0	28.3	3.2	
TDF	0.05	70	0.41	70.7	0.006	
Tenofovir	5	6000	--	--	--	

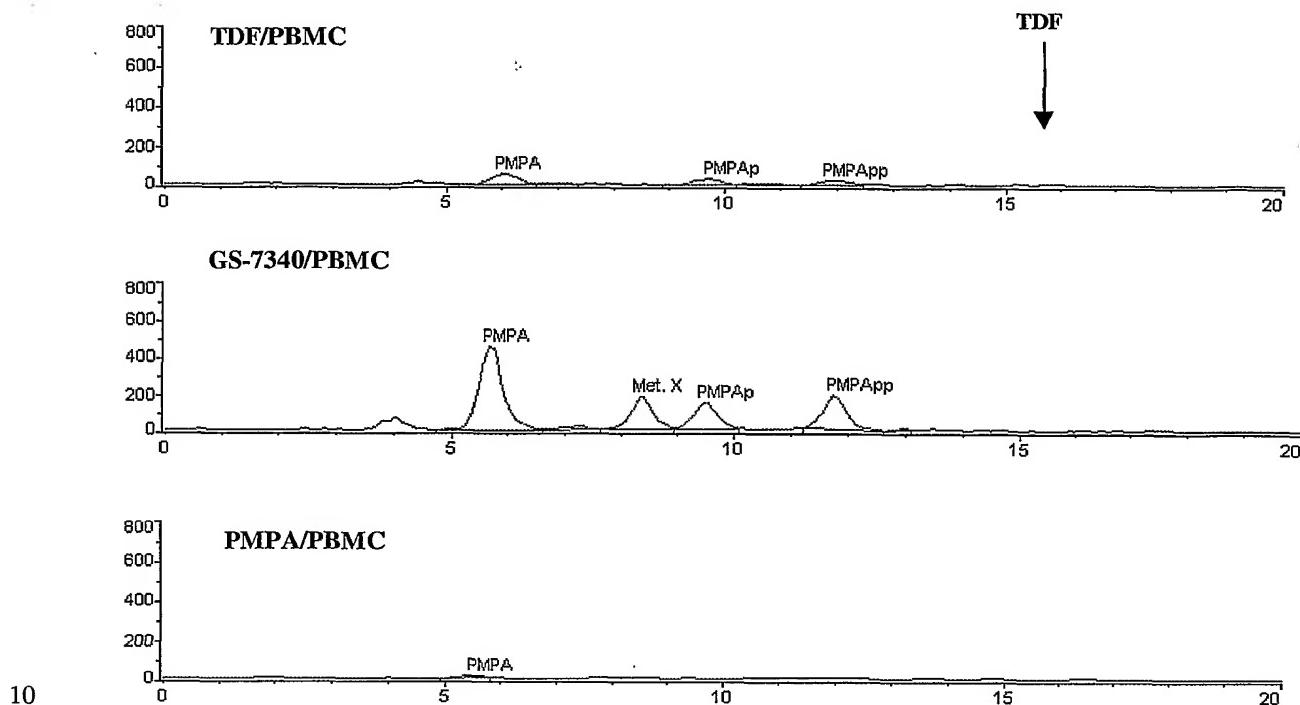
25 In order to estimate the relative intracellular PMPA resulting from the intracellular metabolism of TDF as compared to that from GS-7340, both prodrugs and PMPA were radiolabeled and spiked into intact human whole blood at equimolar concentrations. After 1 hour, plasma, red blood cells (RBCs) and peripheral blood mononuclear cells (PBMCs) were isolated and analyzed by HPLC  
30 with radiometric detection. The results are shown in Table 2.

5 After 1 hour, GS-7340 results in 10x and 30x the total intracellular concentration of PMPA species in PBMCs as compared to TDF and PMPA, respectively. In plasma after 1 hour, 84% of the radioactivity is due to intact GS-  
 10 7340, whereas no TDF is detected at 1 hour. Since no intact TDF is detected in plasma, the 10x difference at 1 hour between TDF and GS-7340 is the minimum difference expected *in vivo*. The HPLC chromatogram for all three compounds in PBMCs is shown in Figure 1.

15 **Table 2. PMPA Metabolites in Plasma, PBMCs and RBCs After 1 h Incubation of PMPA Prodrugs or PMPA in Human Blood.**

Compound	Matrix	Total C-14 Recovered, µg-eq	Metabolites (% of Total Peak Area)					
			PMPA %	PMPAp, %	PMPApp, %	Met. X, %	Met. Y, %	GS 7340, %
GS-7340 (60 µg-eq)	Plasma/FP	43.0	1	-	-	2	13	84
	PBMC	1.25	45	16	21	18	-	-
	RBC/FP	12.6	8	-	-	24	11	57
			PMPA	PMPAp	PMPApp	Mono-POC	GS-4331	
GS-4331 (TDF) (60 µg-eq)	Plasma/FP	48.1	11	-	-	89	-	
	PBMC	0.133	50	25	18	7	-	
	RBC/FP	10.5	93	7.0	-	-	-	
			PMPA	PMPAp	PMPApp			
PMPA (60 µg-eq)	Plasma/FP	55.7	100	-	-			
	PBMC	0.033	86	14	-			
	RBC/FP	3.72	74	10	16			

**Figure 1. HPLC/C-14 Traces of PBMC Extracts from Human Blood Incubated for 1 h at 37°C with TDF, GS-7340 or PMPA.**



Met. X and Met Y (metabolites X and Y) are shown in Table 5. Lower case "p" designates phosphorylation. These results were obtained after 1 hour in human blood. With increasing time, the *in vitro* differences are expected to increase, since 84% of GS-7340 is still intact in plasma after one hour. Because intact GS-7340 is present in plasma after oral administration, the relative clinical efficacy should be related to the IC<sub>50</sub> values seen *in vitro*.

In Table 3 below, IC<sub>50</sub> values of tenofovir, TDF, GS-7340, several nucleosides and the protease inhibitor nelfinavir are listed. As shown, nelfinavir and GS-7340 are 2-3 orders of magnitude more potent than all other nucleotides or nucleosides.

**Table 3. *In Vitro* Anti-HIV-1 Activities of Antiretroviral Compounds**

Compound	IC <sub>50</sub> ( $\mu$ M)
Adefovir (PMEA)	13.4 ± 4.2 <sup>1</sup>
Tenofovir (PMPA)	6.3 ± 3.3 <sup>1</sup>
AZT	0.17 ± 0.08 <sup>1</sup>
3TC	1.8 ± 0.25 <sup>1</sup>
d4T	8 ± 2.5 <sup>1</sup>
Nelfinavir	0.006 ± 0.002 <sup>1</sup>
TDF	0.05
GS 7340	0.005

1. A. S. Mulato and J. M. Cherrington, *Antiviral Research* **36**, 91 (1997)

10 Additional studies of the *in vitro* cell culture anti-HIV-1 activity and CC<sub>50</sub> of separated diastereomers of this invention were conducted and the results tabulated below.

**Table 4. Effect of Diastereomer**

Compound	Diastereomer	$IC_{50}$ ( $\mu M$ )	Fold change	A/B activity	$CC_{50}$ ( $\mu M$ )
<b>PMPA</b>	-	5	1x	-	6000
Ala-methylester	Mixture 1:1	0.025	200x	20x	80
GS-6957a	A	0.0075	670x		
GS-6957b		0.15	33x		
Phe-methylester	Mixture 1:1	0.03	170x	10x	60
GS-7003a	A	0.01	500x		
GS-7003b	B	0.1	50x		
Gly-ethylester	Mixture 1:1	0.5	10x	20x	
GS-7119	A	0.05	100x		>100
GS-7335	B	1.0	5x		
Ala-isopropyl	Mixture 1:1	0.01	500x	12x	
GS-7340	A	0.005	1,000x		40
GS-7339	B	0.06	83x		>100
ABA-ethyl	Mixture 1:1	0.008	625x	7.5x	>100
GS-7342	A	0.004	1,250x		
GS-7341	B	0.03	170x		
Ala-ethyl	Mixture 1:1	0.02	250x	10x	60
GS-7114	A	0.005	1,000x		
GS-7115	B	0.05	100x		

Assay reference: Arimilli, MN, et al., (1997) Synthesis, *in vitro* biological evaluation and oral bioavailability of 9-[2-(phosphonomethoxy)propyl]adenine (PMPA) prodrugs. Antiviral Chemistry and Chemotherapy 8(6):557-564.

"Phe-methylester" is the methylphenylalaninyl monoamidate, phenyl monoester of tenofovir; "gly-methylester" is the methylglycyl monoamidate, phenyl monoester of tenofovir.

In each instance above, isomer A is believed to have the same absolute stereochemistry as GS-7340 (S), and isomer B is believed to have the same absolute stereochemistry that of GS-7339.

5       The *in vitro* metabolism and stability of separated diastereomers were  
determined in PLCE, MT-2 extract and human plasma. A biological sample listed  
below, 80 µL, was transferred into a screw-capped centrifuge tube and incubated at  
37°C for 5 min. A solution containing 0.2 mg/mL of the test compound in a  
suitable buffer, 20 µL, was added to the biological sample and mixed. The reaction  
10 mixture, 20 µL, was immediately sampled and mixed with 60 µL of methanol  
containing 0.015 mg/mL of 2-hydroxymethylnaphthalene as an internal standard  
for HPLC analysis. The sample was taken as the time-zero sample. Then, at  
specific time points, the reaction mixture, 20 µL, was sampled and mixed with 60  
µL of methanol containing the internal standard. The mixture thus obtained was  
15 centrifuged at 15,000 G for 5 min and the supernatant was analyzed with HPLC  
under the conditions described below.

The biological samples evaluated are as follows.

- (1) PLCE (porcine liver carboxyesterase from Sigma, 160 u/mg protein, 21 mg  
20     protein/mL) diluted 20 fold with PBS (phosphated-buffered saline).
- (2) MT-2 cell extract was prepared from MT-2 cells according to the published  
procedure [A. Pompon, I. Lefebvre, J.-L. Imbach, S. Kahn, and D. Farquhar,  
"Antiviral Chemistry & Chemotherapy", 5:91–98 (1994)] except for using HEPES  
buffer described below as the medium.
- 25     (3) Human plasma (pooled normal human plasma from George King Biomedical  
Systems, Inc.)

The buffer systems used in the studies are as follows.

In the study for PLCE, the test compound was dissolved in PBS. PBS (phosphate-  
30     buffered saline, Sigma) contains 0.01 M phosphate, 0.0027 M potassium chloride,  
and 0.137 M sodium chloride. pH 7.4 at 37°C.

In the study for MT-2 cell extracts, the test compound was dissolved in HEPES  
buffer. HEPES buffer contains 0.010 M HEPES, 0.05 M potassium chloride, 0.005 M  
magnesium chloride, and 0.005 M *dl*-dithiothreitol. pH 7.4 at 37°C.

- 5 In the study for human plasma, the test compound was dissolved in TBS. TBS (tris-buffered saline, Sigma) contains 0.05 M Tris, 0.0027 M potassium chloride, and 0.138 M sodium chloride. pH 7.5 at 37°C.

10 The HPLC analysis was carried out under the following conditions.

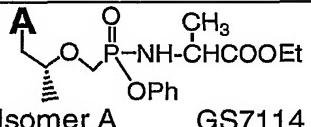
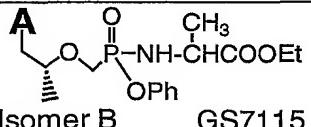
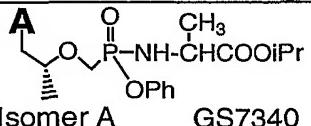
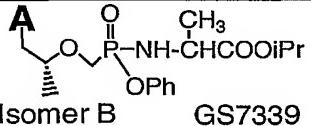
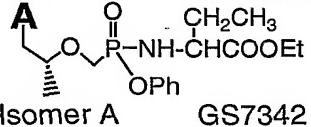
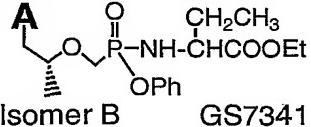
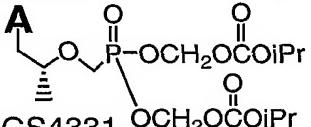
Column: Zorbax Rx-C8, 4.6 x 250 mm, 5  $\mu$   
(MAC-MOD Analytical, Inc. Chadds Ford, PA)  
Detection: UV at 260 nm  
Flow Rate: 1.0 mL/min  
Run Time: 30 min  
Injection Volume: 20  $\mu$ L  
Column Temperature: Ambient temperature

Mobile Phase A: 50 mM potassium phosphate (pH 6.0)/CH<sub>3</sub>CN = 95/5 (v/v)  
Mobile Phase B: 50 mM Potassium phosphate (pH 6.0)/CH<sub>3</sub>CN = 50/50 (v/v)

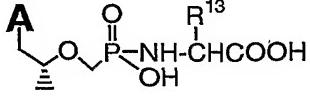
Gradient Run: 0 min 100% Mobile Phase A  
25 min 100% Mobile Phase B  
30 min 100% Mobile Phase B

25 The results are shown below in Table 5 (also including selected IC<sub>50</sub> data from Table 4).

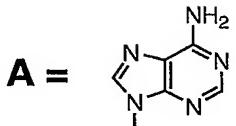
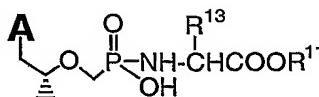
5 Table 5. In Vitro Metabolism of Isomers A and B of PMPA monoamidate at 37°C

No	PMPA monoamidate structure	HIV IC <sub>50</sub> (μM)	PLCE hydrolysis rate and product	MT-2 extract hydrolysis rate and product	Human Plasma Stability (HP)
1	 Isomer A GS7114	0.005	t <sub>1/2</sub> = 2.9 min Met. X & PMPA	t <sub>1/2</sub> = 2.9 min Met. X & PMPA	t <sub>1/2</sub> = 148 min Met. Y
2	 Isomer B GS7115	0.05	t <sub>1/2</sub> = 8.0 min Met. X & PMPA	t <sub>1/2</sub> = 150.6 min Met. X & PMPA	t <sub>1/2</sub> = 495 min Met. Y
3	 Isomer A GS7340	0.005	t <sub>1/2</sub> = 3.3 min Met. X & PMPA	t <sub>1/2</sub> = 28.3 min Met. X & PMPA	t <sub>1/2</sub> = 90.0 min Met. Y
4	 Isomer B GS7339	0.06	t <sub>1/2</sub> = 10.1 min Met. X & PMPA	t <sub>1/2</sub> > 1000 min	t <sub>1/2</sub> = 231 min Met. Y
5	 Isomer A GS7342	0.004	t <sub>1/2</sub> = 3.9 min Met. X	t <sub>1/2</sub> = 49.2 min Met. X & PMPA	t <sub>1/2</sub> = 103 min Met. Y
6	 Isomer B GS7341	0.03	t <sub>1/2</sub> = 11.3 min Met. X	t <sub>1/2</sub> > 1000 min	t <sub>1/2</sub> = 257 min Met. Y
7	 GS4331	0.05	t <sub>1/2</sub> < 0.14 min MonoPOC PMPA	t <sub>1/2</sub> = 70.7 min monoPOC PMPA	t <sub>1/2</sub> = 0.41 min monoPOC PMPA

Met. X:



Met. Y:



5

Example 10

Plasma and PBMC Exposures Following Oral Administration  
Of Prodrug Diastereomers to Beagle Dogs

The pharmacokinetics of GS 7340 were studied in dogs after oral  
10 administration of a 10 mg-eq/kg dose.

**Formulations.** The prodrugs were formulated as solutions in 50 mM citric acid within 0.5 hour prior to dose. All compounds used in the studies were synthesized by Gilead Sciences. The following lots were used:

15

GSI	Amidate Amino acid	AA Ester	Diastereoisomer	Lot Number
GS-7340-2	Alanine	i-Propyl	Isomer A	1504-187-19
GS-7339	Alanine	i-Propyl	Isomer B	1509-185-31
GS7114	Alanine	Ethyl	Isomer A	1509-181-26
GS7115	Alanine	Ethyl	Isomer B	1509-181-22
GS7119	Glycine	Ethyl	Isomer A	1428-163-28
GS7342	$\alpha$ -Aminobutyric Acid	Ethyl	Isomer A	1509-191-12
GS7341	$\alpha$ -Aminobutyric Acid	Ethyl	Isomer B	1509-191-7

**Dose Administration and Sample Collection.** The in-life phase of this study was conducted in accordance with the recommendations of the "Guide for the Care and Use of Laboratory Animals" (National Institutes of Health publication 86-23) and 20 was approved by an Institutional Animal Care and Use Committee. Fasted male beagle dogs ( $10 \pm 2$  kg) were used for the studies. Each drug was administered as a single dose by oral gavage (1.5–2 ml/kg). The dose was 10 mg-equivalent of PMPA/kg. For PBMCs, blood samples were collected at 0 (pre-dose), 2, 8, and 24 h post-dose. For plasma, blood samples were collected at 0 (pre-dose), 5, 15, and 30 25 min, and 1, 2, 3, 4, 6, 8, 12 and 24 h post-dose. Blood (1.0 ml) was processed immediately for plasma by centrifugation at 2,000 rpm for 10 min. Plasma samples were frozen and maintained at 70°C until analyzed.

**Peripheral Blood Mononuclear Cell (PBMC) preparation.** Whole blood (8 ml) 30 drawn at specified time points was mixed in equal proportion with phosphate buffered saline (PBS), layered onto 15 ml of Ficoll-Paque solution (Pharmacia Biotech,) and centrifuged at 400  $\times$  g for 40 min. PBMC layer was removed and

washed once with PBS. Formed PMBC pellet was reconstituted in 0.5 ml of PBS, cells were resuspended, counted using hemocytometer and maintained at 70°C until analyzed. The number of cells multiplied by the mean single-cell volume was used in calculation of intracellular concentrations. A reported value of 200 femtoliters/cell was used as the resting PBMC volume (B. L. Robins, R.V. Srinivas, C. Kim, N. Bischofberger, and A. Fridland, *Antimicrob. Agents Chemother.* 42, 612 (1998)).

**Determination of PMPA and Prodrugs in plasma and PBMCs.** The concentration of PMPA in dog plasma samples was determined by derivatizing PMPA with chloroacetaldehyde to yield a highly fluorescent N<sup>1</sup>, N<sup>6</sup>-ethenoadenine derivative (L. Naesens, J. Balzarini, and E. De Clercq, *Clin. Chem.* 38, 480 (1992)). Briefly, plasma (100 µl) was mixed with 200 µl acetonitrile to precipitate protein. Samples were then evaporated to dryness under reduced pressure at room temperature. Dried samples were reconstituted in 200 µl derivatization cocktail (0.34% chloroacetaldehyde in 100 mM sodium acetate, pH 4.5), vortexed, and centrifuged. Supernatant was then transferred to a clean screw-cap tube and incubated at 95°C for 40 min. Derivatized samples were then evaporated to dryness and reconstituted in 100 µl of water for HPLC analysis.

Before intracellular PMPA could be determined by HPLC, the large amounts of adenine related ribonucleotides present in the PBMC extracts had to be removed by selective oxidation. We used a modified procedure of Tanaka et al (K. Tanaka, A. Yoshioka, S. Tanaka, and Y. Wataya, *Anal. Biochem.*, 139, 35 (1984)). Briefly, PBMC samples were mixed 1:2 with methanol and evaporated to dryness under reduced pressure. The dried samples were derivatized as described in the plasma assay. The derivatized samples were mixed with 20 µL of 1M rhamnose and 30 µL of 0.1M sodium periodate and incubated at 37°C for 5 min. Following incubation, 40 µL of 4M methylamine and 20 µL of 0.5M inosine were added. After incubation at 37°C for 30 min, samples were evaporated to dryness under reduced pressure and reconstituted in water for HPLC analysis.

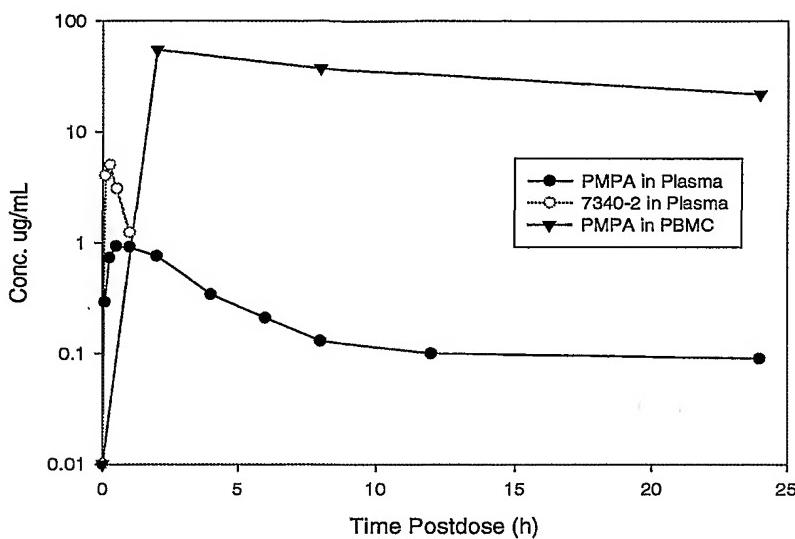
5 No intact prodrug was detected in any PBMC samples. For plasma samples potentially containing intact prodrugs, experiments were performed to verify that no further conversion to PMPA occurred during derivatization. Prodrug standards were added to drug-free plasma and derivatized as described. There were no detectable levels of PMPA present in any of the plasma samples, and the projected  
10 % of conversion was less than 1%.

The HPLC system was comprised of a P4000 solvent delivery system with AS3000 autoinjector and F2000 fluorescence detector (Thermo Separation, San Jose, CA). The column was an Inertsil ODS-2 column (4.6 × 150 mm). The mobile phases used  
15 were: A, 5% acetonitrile in 25 mM potassium phosphate buffer with 5mM tetrabutyl ammonium bromide (TBABr), pH 6.0; B, 60% acetonitrile in 25 mM potassium phosphate buffer with 5 mM TBABr, pH 6.0. The flow rate was 2 ml/min and the column temperature was maintained at 35°C by a column oven. The gradient profile was 90% A/10% B for 10 min for PMPA and 65%A/35%B for 10 min for the  
20 prodrug. Detection was by fluorescence with excitation at 236 nm and emission at 420 nm, and the injection volume was 10 µl. Data was acquired and stored by a laboratory data acquisition system (PeakPro, Beckman, Allendale, NJ).

**Pharmacokinetic Calculations.** PMPA and prodrug exposures were expressed as  
25 areas under concentration curves in plasma or PBMC from zero to 24 hours (AUC). The AUC values were calculated using the trapezoidal rule.

**Plasma and PBMC Concentrations.** The results of this study is shown in Figures 2 and 3. Figure 2 shows the time course of GS 7340-2 metabolism summary of plasma and PBMC exposures following oral administration of pure diastereoisomers of the  
30 PMPA prodrugs.

5      **Figure 2. PMPA and Prodrug Concentration in Plasma and PBMCs Following Oral Administration of GS 7340-2 to Dogs at 10 mg-eq/kg.**

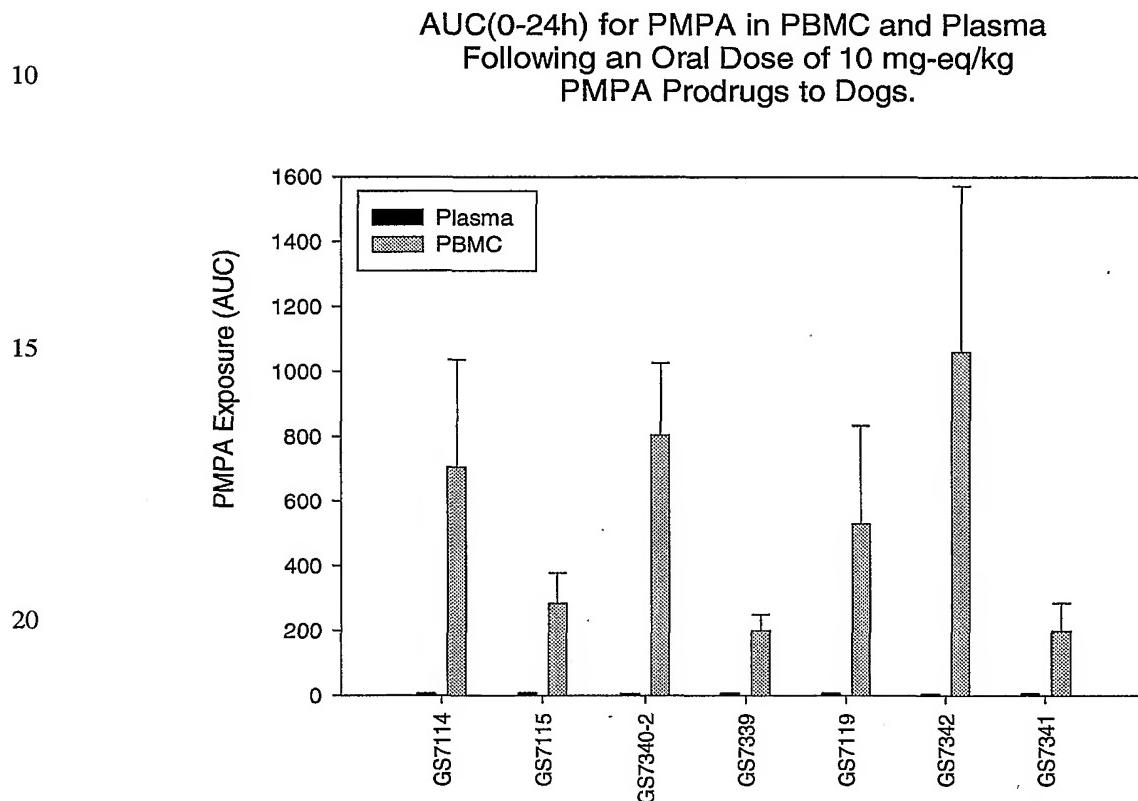


10

The bar graph in Figure 2 shows the AUC (0-24h) for tenofovir in dog PBMCs and plasma after administration of PMPA s.c., TDF and amidate ester prodrugs. All of the amidate prodrugs exhibited increases in PBMC exposure. For example, GS 7340 results in a ~21-fold increase in PBMC exposure as compared to PMPA s.c. and TDF; and a 6.25-fold and 1.29-fold decrease in plasma exposure, respectively.

15

5      **Figure 3. Depicts Tenofovir Exposure in PBMCs and Plasma Upon  
Administration of 10 mg-eq/kg in dogs**



25

These data establish *in vivo* that GS 7340 can be delivered orally, minimizes systemic exposure to PMPA and greatly enhances the intracellular concentration of PMPA in the cells primarily responsible for HIV replication.

30

Table 6

PMPA Exposure in PBMC and Plasma from Oral Prodrugs of PMPA in Dogs

GS#	Moiety	PMPA AUC in Plasma			PMPA AUC in PBMC			Prodrug in Plasma	PBMC/Plasma Exposure Ratio
		Mean	StDev	N	Mean	StDev	N		
GS-7114	Mono-Ala-Et-A	5.8	0.9	2	706	331	5	YES	122
GS-7115	Mono-Ala-Et-B	6.6	1.5	2	284	94	5	YES	43
GS-7340-2	Mono-Ala-iPr-A	5.0	1.1	5	805	222	5	YES	161
GS-7339	Mono-Ala-iPr-A	6.4	1.3	2	200	57	5	YES	31
GS-7119	Mono-Gly-Et-A	6.11	1.86	2	530	304	5	YES	87
GS-7342	Mono-ABA-Et-A	4.6	1.2	2	1060	511	5	YES	230
GS7341	Mono-ABA-Et-B	5.8	1.4	2	199	86	5	YES	34

Example 11

20

## Biodistribution of GS-7340

As part of the preclinical characterization of GS-7340, its biodistribution in dogs was determined. The tissue distribution of GS-7340 (isopropyl alaninyl monoamide, phenyl monoester of tenofovir) was examined following oral 25 administration to beagle dogs. Two male animals were dosed orally with <sup>14</sup>C=GS-7340 (8.85 mg-equiv. of PMPA/kg, 33.2 µCi/kg; the 8-carbon of adenine is labeled) in an aqueous solution (50 mM citric acid, pH 2.2). Plasma and peripheral blood mononuclear cells (PBMCs) were obtained over the 24-hr period. Urine and feces 30 were cage collected over 24 hr. At 24 h after the dose, the animals were sacrificed and tissues removed for analysis. Total radioactivity in tissues was determined by oxidation and liquid scintillation counting.

The biodistribution of PMPA after 24 hours after a single oral dose of radiolabelled GS 7340 is shown in Table 4 along with the data from a previous study with TDF (GS-4331). In the case of TDF, the prodrug concentration in the plasma is below the level of assay detection, and the main species observed in 35 plasma is the parent drug. Levels of PMPA in the lymphatic tissues, bone marrow, and skeletal muscle are increased 10-fold after administration of GS-7340.

5           Accumulation in lymphatic tissues is consistent with the data observed from  
the PBMC analyses, since these tissues are composed primarily of lymphocytes.  
Likewise, accumulation in bone marrow is probably due to the high percentage of  
lymphocytes (70%) in this tissue.

10          Table 7. Excretion and Tissue Distribution of Radiolabelled GS-7340 in Dogs (Mean,  
N=2) Following an Oral Dose at 10 mg-eq. PMPA/kg.

Tissue/Fluid	GS-4331		GS-7340		Tissue Conc. Ratio of GS 7340 to GS-4331
	% Dose	Conc. (ug-eq/g)	% Dose	Conc. (ug-eq/g)	
Liver	12.40	38.30	16.45	52.94	1.4
Kidney	4.58	87.90	3.78	80.21	0.9
Lungs	0.03	0.53	0.34	4.33	8.2
Iliac Lymph Nodes	0.00	0.51	0.01	5.42	10.6
Axillary Lymph Nodes	0.00	0.37	0.01	5.54	14.8
Inguinal Lymph Nodes	0.00	0.28	0.00	4.12	15.0
Mesenteric Lymph Nodes	0.00	1.20	0.04	6.88	5.7
Thyroid Gland	0.00	0.30	0.00	4.78	15.8
Pituitary Gland	0.00	0.23	0.00	1.80	7.8
Salivary Gland (L+R)	0.00	0.45	0.03	5.54	12.3
Adrenal Gland	0.00	1.90	0.00	3.47	1.8
Spleen	0.00	0.63	0.17	8.13	12.8
Pancreas	0.00	0.57	0.01	3.51	6.2
Prostate	0.00	0.23	0.00	2.14	9.1
Testes (L+R)	0.02	1.95	0.02	2.01	1.0
Skeletal Muscle	0.00	0.11	0.01	1.12	10.1
Heart	0.03	0.46	0.15	1.97	4.3
Femoral Bone	0.00	0.08	0.00	0.28	3.5
Bone Marrow	0.00	0.20	0.00	2.05	10.2
Skin	0.00	0.13	0.00	0.95	7.2
Abdominal fat	0.00	0.16	0.00	0.90	5.8
Eye (L+R)	0.00	0.06	0.00	0.23	3.7
Brain	0.00	<LOD	0.00	<LOD	n.d.
Cerebrospinal Fluid	0.00	<LOD	0.00	0.00	n.d.
Spinal Cord	0.00	<LOD	0.00	0.04	n.d.
Stomach	0.11	1.92	0.26	2.68	1.4
Jejunum	1.34	3.01	0.79	4.16	1.4
Duodenum	0.49	4.96	0.44	8.77	1.8
Ileum	0.01	0.50	0.16	4.61	9.2
Large Intestine	1.63	5.97	2.65	47.20	7.9
Gall bladder	0.00	3.58	0.04	25.02	7.0
Bile	0.00	9.63	0.22	40.48	4.2
Feces	40.96	n.d.	0.19	n.d.	n.a.
Total GI Tract Contents	5.61	n.d.	21.64	n.d.	n.a.
Urine	23.72	n.d.	14.73	n.d.	n.a.
Plasma at 24 h	0.00	0.20	0.00	0.20	1.0
Plasma at 0.25 h	n.a.	3.68	n.a.	3.48	0.9
PBMC*	0.00	n.d.	0.00	63.20	n.d.
Whole Blood	0.00	0.85	0.16	0.20	0.2
<b>Total Recovery</b>	<b>81.10</b>		<b>68.96</b>		

\* Calculated using typical recovery of  $15 \times 10^6$  cells total, and mean PBMC volume of 0.2 picoliters/cell

n.s. = no sample, n.a. = not applicable, n.d. = not determined.

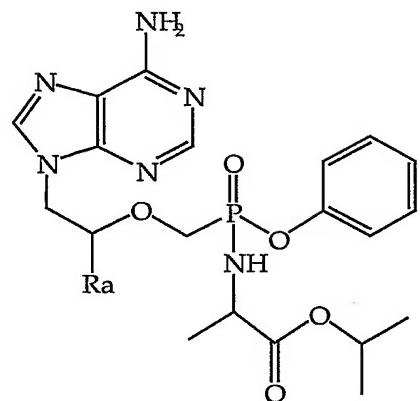
5        CLAIMS:

1.        A screening method for identifying a methoxyphosphonate nucleotide analogue prodrug conferring enhanced activity in a target tissue comprising:
  - (a)      providing at least one of said prodrugs;
  - (b)      selecting at least one therapeutic target tissue and at least one non-target tissue;
  - (c)      administering the prodrug to the target tissue and to said at least one non-target tissue; and
  - (d)      determining the relative activity conferred by the prodrug in the tissues in step (c).
2.        The method of claim 1 wherein the activity is antiviral activity or antitumor activity.
- 20      3.        The method of claim 2 wherein the activity is antiviral activity.
4.        The method of claim 3 wherein the activity is anti-HIV or anti-HBV activity.
5.        The method of claim 1 wherein the prodrug is a prodrug of PMPA or PMEA.
- 25      6.        The method of claim 5 wherein the prodrug is a phosphonoamidate, phosphonoester or mixed phosphonoamidate/phosphonoester.
7.        The method of claim 6 wherein the amidate is an amino acid amide.
- 30      8.        The method of claim 6 wherein the ester is an aryl ester.
9.        The method of claim 1 further comprising selecting a prodrug having a relative activity in the target tissue that is greater than 10 times that of the non-target tissue.

- 5    10.    The method of claim 1 wherein the target and non-target tissue are in an animal, the prodrug is administered to the animal and the relative activity is determined by analysis of the animal tissues after administration of the prodrug.
- 10    11.    The method of claim 1 wherein activity in the target and non-target tissues is determined by assaying the amount of at least one metabolite of the prodrug in the tissues.
- 15    12.    The method of claim 12 wherein the metabolite is the parental drug.
- 15    13.    The method of claim 12 wherein the metabolite is the diphosphate of the parental drug.
- 20    14.    The method of claim 1 wherein the target tissue is virally infected tissue and the non-target tissue is the same tissue which is not virally infected.
- 20    15.    The method of claim 1 wherein the target tissue is lymphoid tissue and the activity is anti-HIV activity.
- 25    16.    The method of claim 1 wherein the target tissue is liver and the activity is anti-HBV activity.
- 25    17.    The method of claim 1 wherein the target tissue is hematological and the activity is antitumor activity.
- 30    18.    The method of claim 1 wherein the target tissue is malignant and the non-target tissue is the same tissue but non-malignant.

5 19. A compound having the structure (1)

10



(1)

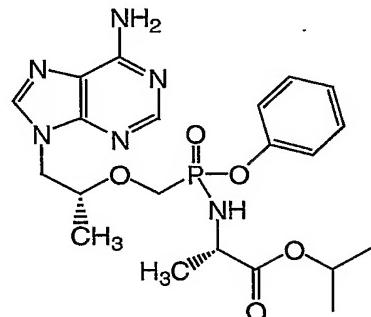
15 where Ra is H or methyl,

and chirally enriched compositions thereof, salts, their free base and solvates thereof.

20 20. A compound having the structure (2)

20

25

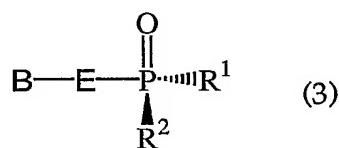


(2)

and its enriched diasteromers, salts, free base and solvates.

30 21. A diastereomerically enriched compound having the structure (3)

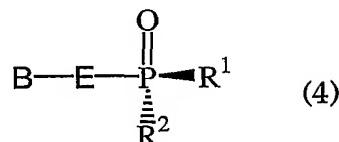
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5

which is substantially free of the diastereomer (4)

10



wherein

$R^1$  is an oxyester which is hydrolyzable *in vivo*, or hydroxyl;

B is a heterocyclic base;

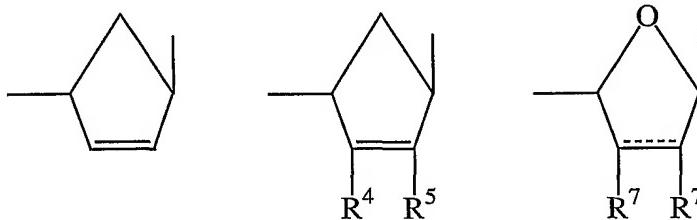
15  $R^2$  is hydroxyl, or the residue of an amino acid bonded to the P atom

through an amino group of the amino acid and having each carboxy substituent of the amino acid optionally esterified, but not both of  $R^1$  and  $R^2$  are hydroxyl;

E is  $-(CH_2)_2-$ ,  $-CH(CH_3)CH_2-$ ,  $-CH(CH_2F)CH_2-$ ,  $-CH(CH_2OH)CH_2-$ ,  
 $-CH(CH=CH_2)CH_2-$ ,  $-CH(C\equiv CH)CH_2-$ ,  $-CH(CH_2N_3)CH_2-$ ,

20

25



$-CH(R^6)OCH(R^{6'})-$ ,  $-CH(R^9)CH_2O-$  or  $-CH(R^8)O-$ , wherein the right hand bond is linked to the heterocyclic base;

the broken line represents an optional double bond;

30  $R^4$  and  $R^5$  are independently hydrogen, hydroxy, halo, amino or a

substituent having 1-5 carbon atoms selected from acyloxy, alkyoxy, alkylthio, alkylamino and dialkylamino;

$R^6$  and  $R^{6'}$  are independently H,  $C_1-C_6$  alkyl,  $C_1-C_6$  hydroxyalkyl, or  $C_2-C_7$  alkanoyl;

35  $R^7$  is independently H,  $C_1-C_6$  alkyl, or are taken together to form  $-O-$  or  
 $-CH_2-$ ;

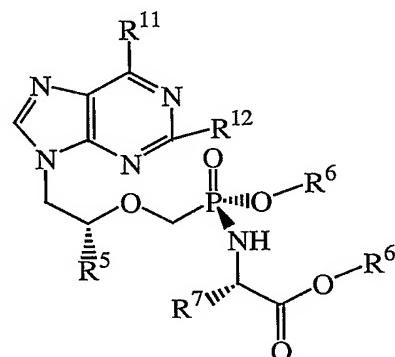
5        R<sup>8</sup> is H, C<sub>1</sub>-C<sub>6</sub> alkyl, C<sub>1</sub>-C<sub>6</sub> hydroxyalkyl or C<sub>1</sub>-C<sub>6</sub> haloalkyl; and

R<sup>9</sup> is H, hydroxymethyl or acyloxymethyl;

and their salts, free base, and solvates.

22.      A diastereomerically enriched compound having the structure (5a)

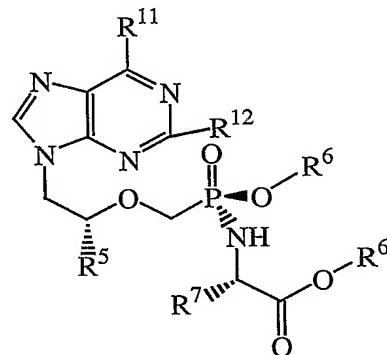
10



(5a)

which is substantially free of diastereomer (5b)

20



(5b)

wherein

30        R<sup>5</sup> is methyl or hydrogen;

R<sup>6</sup> independently is H, alkyl, alkenyl, alkynyl, aryl or arylalkyl, or R<sup>6</sup> independently is alkyl, alkenyl, alkynyl, aryl or arylalkyl which is substituted with from 1 to 3 substituents selected from alkylamino, alkylaminoalkyl, dialkylaminoalkyl, dialkylamino, hydroxyl, oxo, halo, amino, alkylthio, alkoxy,

5 alkoxyalkyl, aryloxy, aryloxyalkyl, arylalkoxy, arylalkoxyalkyl, haloalkyl, nitro, nitroalkyl, azido, azidoalkyl, alkylacyl, alkylacylalkyl, carboxyl, or alkylacylamino;

R<sup>7</sup> is the side chain of any naturally-occurring or pharmaceutically acceptable amino acid and which, if the side chain comprises carboxyl, the carboxyl group is optionally esterified with an alkyl or aryl group;

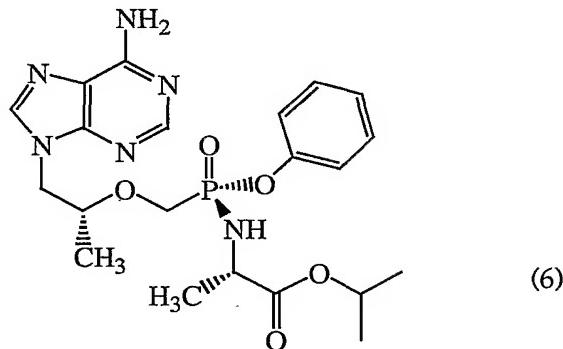
10 R<sup>11</sup> is amino, alkylamino, oxo, or dialkylamino; and

R<sup>12</sup> is amino or H;

and its salts, tautomers, free base and solvates.

23. A compound of structure (6)

15



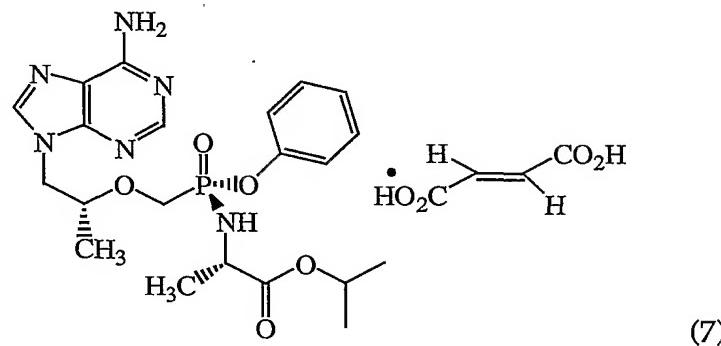
20

and its salts and solvates.

25

24. A compound of structure (7)

30



35

- 5     25. A composition comprising a compound of any of claims 19-24 and a pharmaceutically effective excipient.
26. The composition of claim 25 wherein the excipient is a gel.
- 10    27. The composition of claim 25 which is suitable for topical administration.
28. A method for antiviral therapy or prophylaxis comprising administering a compound of any of claims 19-24 in a therapeutically or prophylactically effective amount to a subject in need of such therapy or prophylaxis.
- 15    29. A method for use of magnesium alkoxide comprising reacting 9-(2-hydroxypropyl)adenine (HPA) or 9-(2-hydroxyethyl)adenine (HEA), magnesium alkoxide, and protected *p*-toluenesulfonyloxymethylphosphonate.
- 20    30. The method of claim 29 further comprising recovering PMPA or PMEA, respectively.
31. The method of claim 29 wherein the phosphonate of the *p*-toluenesulfonyloxymethylphosphonate is protected by ethyl ester.
- 25    32. The method of claim 29 wherein the alkoxide is a C<sub>1</sub>-C<sub>6</sub> alkoxide.
33. The method of claim 32 wherein the alkoxide is t-butyl or isopropyl oxide.

(19) World Intellectual Property Organization  
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(10) International Publication Number  
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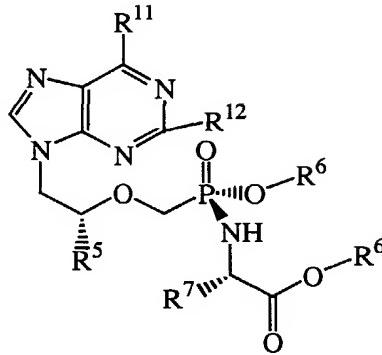
(71) Applicant (for all designated States except US): **GILEAD SCIENCES, INC.** [US/US]; 333 Lakeside Drive, Foster City, CA 94404 (US).

(72) Inventors; and

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(54) Title: PRODRUGS OF PHOSPHONATE NUCLEOTIDE ANALOGUES AND METHODS FOR SELECTING AND MAKING SAME

**WO 02/008241 A3**



(57) Abstract: A novel method is provided for screening prodrugs of methoxyphosphonate nucleotide analogues to identify prodrugs selectively targeting desired tissues with antiviral or antitumor activity. This method has led to the identification of novel mixed ester-amidates of PMPA for retroviral or hepadnaviral therapy, including compounds of structure (5a) having substituent groups as defined herein. Compositions of these novel compounds in pharmaceutically acceptable excipients and their use in therapy and prophylaxis are provided. Also provided is an improved method for the use of magnesium alkoxide for the preparation of starting materials and compounds for use herein.

**Published:**

- with international search report
- before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments

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## INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 01/23104

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C07F9/6561 C07H19/20 G01N33/53 A61K31/675 A61P31/12  
A61P35/00

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C07F C07H G01N A61K A61P

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, CHEM ABS Data

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 5 977 089 A (OLIYAI REZA ET AL) 2 November 1999 (1999-11-02) examples -----	1,18
A	EP 0 481 214 A (SQUIBB BRISTOL MYERS CO) 22 April 1992 (1992-04-22) the whole document ----- -/-	1,18



Further documents are listed in the continuation of box C.



Patent family members are listed in annex.

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Date of the actual completion of the international search

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## INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 01/23104

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	SIDDQUI A Q ET AL: "DESIGN AND SYNTHESIS OF LIPOPHILIC PHOSPHORAMIDATE D4T-MP PRODRUGS EXPRESSING HIGH POTENCY AGAINST HIV IN CELL CULTURE: STRUCTURAL DETERMINANTS FOR IN VITRO ACTIVITY AND QSAR" JOURNAL OF MEDICINAL CHEMISTRY, AMERICAN CHEMICAL SOCIETY. WASHINGTON, US, vol. 42, no. 20, 1999, pages 4122-4128, XP000999173 ISSN: 0022-2623 page 4122 -----	1,18
A	WO 96 37503 A (GENTA INC) 28 November 1996 (1996-11-28) figure 9 -----	21
P,X	H. CHAPMAN ET AL.: "Practical synthesis, separation, and stereochemical assignment of the PMPA pro-drug GS-7340" NUCLEOSIDES, NUCLEOTIDES & NUCLEIC ACIDS, vol. 20, 2001, pages 621-628, XP002202043 the whole document -----	1-33

## INTERNATIONAL SEARCH REPORT

International application No.  
PCT/US 01/23104

### Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)

This International Search Report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1.  Claims Nos.: because they relate to subject matter not required to be searched by this Authority, namely:  

Although claim 28 is directed to a method of treatment of the human/animal body, the search has been carried out and based on the alleged effects of the compound.
2.  Claims Nos.: because they relate to parts of the International Application that do not comply with the prescribed requirements to such an extent that no meaningful International Search can be carried out, specifically:
3.  Claims Nos.: because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

### Box II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

1.  As all required additional search fees were timely paid by the applicant, this International Search Report covers all searchable claims.
2.  As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3.  As only some of the required additional search fees were timely paid by the applicant, this International Search Report covers only those claims for which fees were paid, specifically claims Nos.:
4.  No required additional search fees were timely paid by the applicant. Consequently, this International Search Report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

#### Remark on Protest

- The additional search fees were accompanied by the applicant's protest.  
 No protest accompanied the payment of additional search fees.

## INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 01/23104

Patent document cited in search report		Publication date		Patent family member(s)	Publication date
US 5977089	A	02-11-1999	US US US AU JP TW WO	5922695 A 6069249 A 6043230 A 3897097 A 2000515866 T 444020 B 9804569 A1	13-07-1999 30-05-2000 28-03-2000 20-02-1998 28-11-2000 01-07-2001 05-02-1998
EP 0481214	A	22-04-1992	AT BR CA DE DE DK EP ES HK JP JP US US	167679 T 1100467 A3 2051239 A1 69129650 D1 69129650 T2 481214 T3 0481214 A1 2118069 T3 1010053 A1 3164385 B2 4230694 A 5663159 A 5792756 A	15-07-1998 06-06-2000 15-03-1992 30-07-1998 25-03-1999 22-02-1999 22-04-1992 16-09-1998 11-06-1999 08-05-2001 19-08-1992 02-09-1997 11-08-1998
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